

ISOCS Short Course Winter 2023

16 - 19 January 2023 – Bormio, Italy

ENVIRONMENTAL GAS & ODOUR SENSING

in collaboration with EU project

→ **SENSOFT** ←

Develop your knowledge of environmental gas & odour sensing at the ISOCS Short Course in Bormio, Italy

This Short Course will cover topics including:

- Environmental odour monitoring
- Environmental data analysis and processing
- Advanced applications with drones
- Trace detection of air pollutants
- Wireless gas sensing networks
- Sensor calibration techniques

Speakers including:

- Prof. Saverio De Vito, ENEA; President ISOCS
- Prof. Laura Capelli, Politecnico di Milano
- Prof. Agustin Gutierrez Galvez, University of Barcelona
- Prof. Eduard Llobet Valero, Universitat Rovira I Virgili
- Prof. Achim J. Lilienthal, Örebro University
- Dr. Jan Mitrovics, JLM Innovation GmbH, Germany

SENSOFT: Smart Sensing for Rapid Response to chemical threats on soft targets

<http://deeea.urv.cat/sensoft/>



*SENSOFT-VOGAS- ISOCS Meeting See (Austria), 27-30 Jan 2020
From nanomaterials for sensing and preconcentration to trace level detection applications*

SERS for Sensing in Gas Phase

Dr. Maria Pilar Pina

Instituto de Nanociencia y Materiales de Aragon (Spain)

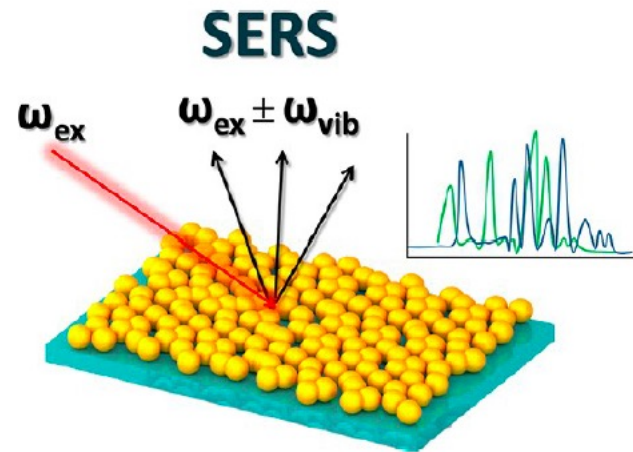
mapina@unizar.es

orcid.org/0000-0001-9897-6527

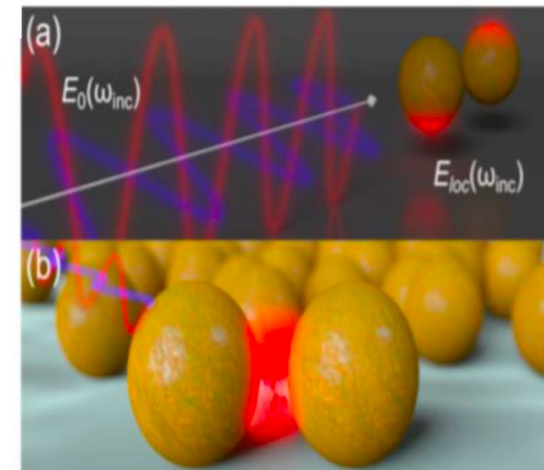
SERS stands for...

1. Ship Emergency Response Service
2. State Employees' Retirement System
3. Software Engineering Release System
4. ?

- Absorption
- Vibrational Excitation
- Re-emission



ACS Nano 2020, 14, 28-117



OUTLINE

- 1. Chemical Sensing & Scenario Driven Requirements**
- 2. SERS Fundamentals**
 - **Spectroscopy & Raman Spectroscopy**
 - **Surface Enhanced Raman Spectroscopy: SERS Effect & substrates**
 - **Measuring SERS**
 - **Portable Instrumentation**
 - **Application Fields**
- 3. SERS for Gas Sensing**
 - **Challenges & Strategies**
 - **Our Approach**
 - **Future Work**
- 4. Take-Home Messages**

1. Chemical Sensing & Scenario Driven Applications



Security and Public Safety

providing real-time monitoring and early detection



Healthcare and Life Science

helping to diagnose and access general health



Home Sensing and Automation

providing safety and security within your home

Reliability

Robustness

KPIs

Response time

Selectivity

Ultra sensitivity

Cost

User friendly

Portability



Agriculture

providing valuable information throughout the growth cycle



Manufacturing

providing consistency within manufacturing



Industrial

providing optimization, monitoring and environmental compliance



Personal Health

monitor exercise, sleep, and health information

Social Challenge: Secure Societies & Chemical Threats

Smart **Sensing** for Rapid Response to chemical threats on **soft** targets (SENSOFT)

Exploiting Surface Enhanced Raman Spectroscopy (**SERS**) and Advanced Algorithms For Guiding Responses to Potential Chemical Threats (SERSING)

... From Odours to CWAs* & TICs



**A Clear Example...Nerve Agents (NAs) Detection*



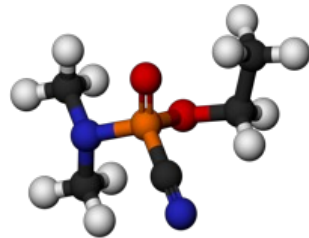
This work has received funding from the European Union's Horizon 2020 research and innovation programme under grant agreements N° 823895 & N° 883390.

Social Challenge: Secure Societies & Chemical Threats

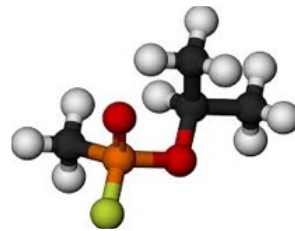
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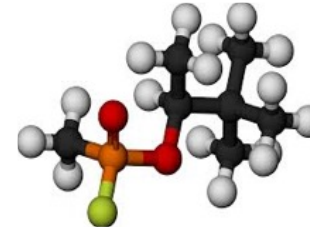
A Clear Example...Nerve Agents (NAs) Detection*



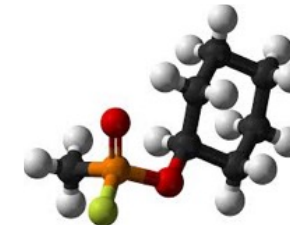
Tabun



SARIN



Soman



Cyclosarin

Phosphorous P
Nitrogen N
Fluorine F
Oxygen O
Carbon C
Hydrogen H

| | Tabun | Sarin | Soman | Cyclosarin |
|--|--|--|--|--|
| <i>NATO designation</i> | GA | GB | GD | GF |
| <i>Chemical formula</i> | C ₅ H ₁₁ N ₂ O ₂ P | C ₄ H ₁₀ FO ₂ P | C ₇ H ₁₆ FO ₂ P | C ₇ H ₁₄ FO ₂ P |
| <i>Appearance</i> | Colorless to brown liquid | Colorless liquid | Colorless liquid | Colorless liquid |
| <i>Odor</i> | None | None | Fruity | None |
| <i>Vapor density (relative to air)</i> | 5.6 | 4.9 | 6.3 | 6.2 |
| <i>Vapor pressure (@ 20°C)</i> | 5 Pa | 280 Pa | 53 Pa | 6 Pa |

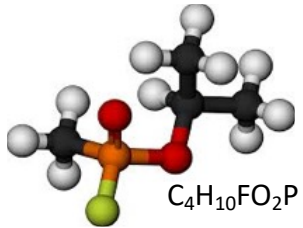
Data extracted from www.epa.gov

*Common Surrogate: DMMP Dimethyl methyl phosphonate CAS Number: 756-79-6

Social Challenge: Secure Societies & Chemical Threats

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Exploiting Surface Enhanced Raman Spectroscopy (**SERS**) and Advanced Algorithms For Guiding Responses to Potential Chemical Threats (SERSING)



Phosphorous P
Nitrogen N
Fluorine F
Oxygen O
Carbon C
Hydrogen H

SARIN

- Accute toxicity.....**SHORT RESPONSE TIME**
- Reliable response.....**LOW FALSE ALARM RATE**
- **Trace detection**.....**SENSITIVITY @ ppt-ppb level**
- Presence of **interferences**.....**ON FIELD USE**

Discomfort

Impaired ability to escape

Life-threatening Effects

| Effect | SARIN exposure time | | | | |
|-------------------------------|---|---|---|---|---|
| | 10 min | 30 min | 60 min | 240 min | 480 min |
| AEGL-1 <i>Nondisabling</i> | 1.2 ppb (6.9 $\mu\text{g}/\text{m}^3$) | 0.68 ppb (4.0 $\mu\text{g}/\text{m}^3$) | 0.48 ppb (2.8 $\mu\text{g}/\text{m}^3$) | 0.24 ppb (1.4 $\mu\text{g}/\text{m}^3$) | 0.17 ppb (1.0 $\mu\text{g}/\text{m}^3$) |
| AEGL-2 <i>Disabling</i> | 15 ppb (87 $\mu\text{g}/\text{m}^3$) | 8.5 ppb (50 $\mu\text{g}/\text{m}^3$) | 6.0 ppb (35 $\mu\text{g}/\text{m}^3$) | 2.9 ppb (17 $\mu\text{g}/\text{m}^3$) | 2.2 ppb (13 $\mu\text{g}/\text{m}^3$) |
| AEGL-3 <i>Lethal</i> | 64 ppb (380 $\mu\text{g}/\text{m}^3$) | 32 ppb (190 $\mu\text{g}/\text{m}^3$) | 22 ppb (130 $\mu\text{g}/\text{m}^3$) | 12 ppb (70 $\mu\text{g}/\text{m}^3$) | 8.7 ppb (51 $\mu\text{g}/\text{m}^3$) |

AEGL: Acute Exposure Guideline Levels

Data extracted from www.epa.gov

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Exploiting Surface Enhanced Raman Spectroscopy (**SERS**) and Advanced Algorithms For Guiding Responses to Potential Chemical Threats (SERSING)



International Forum to Advance
FIRST RESPONDER INNOVATION

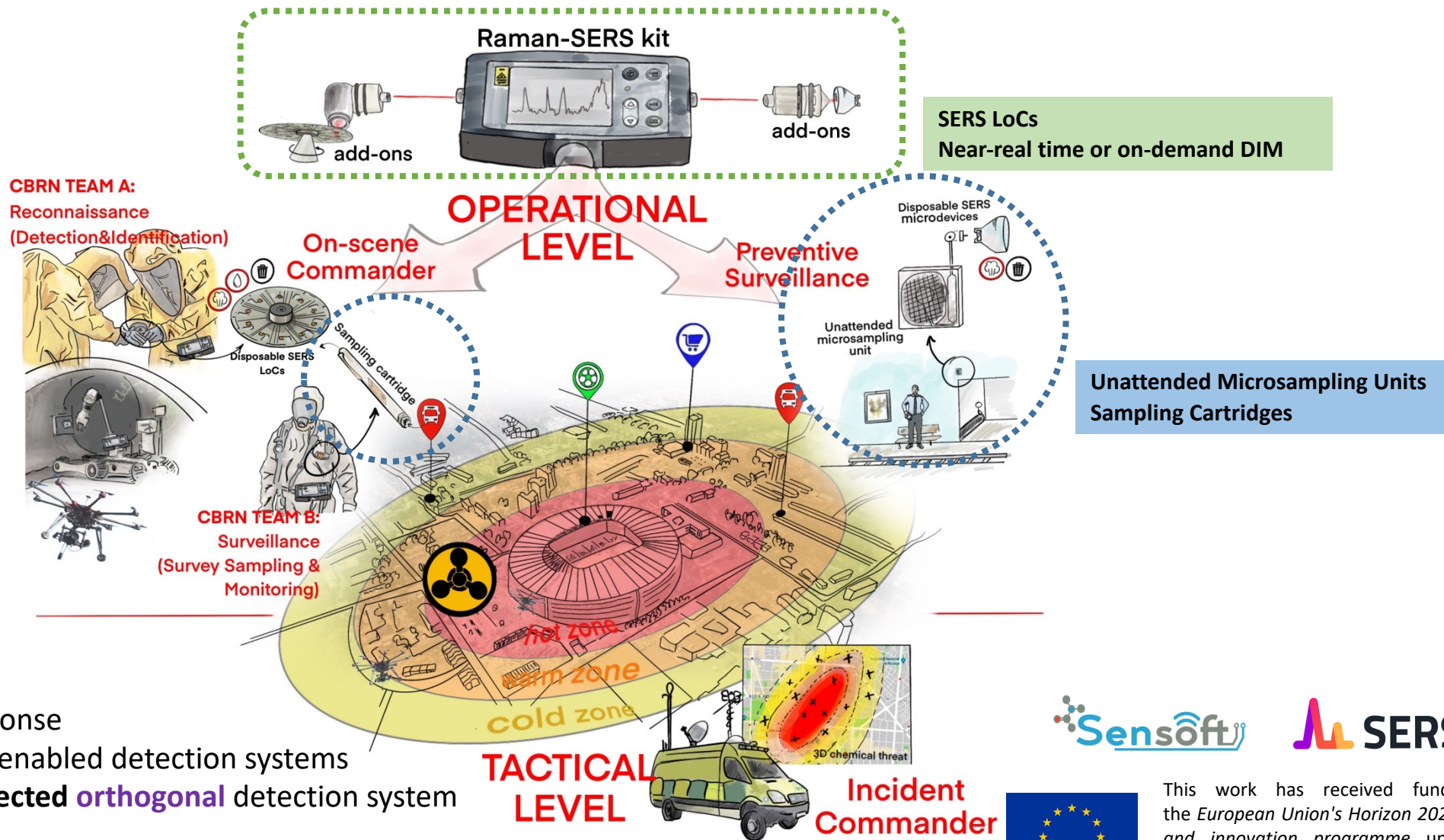
Lack of **miniature, fieldable** and **affordable** tools and systems for **detection, identification** and **monitoring (DIM)** of **Chemical Threats** (**AEGLs @ ppb level**)



Social Challenge: Secure Societies & Chemical Threats

Smart **Sensing** for Rapid Response to chemical threats on **soft** targets (SENSOFT)

Exploiting Surface Enhanced Raman Spectroscopy (**SERS**) and Advanced Algorithms For Guiding Responses to Potential Chemical Threats (SERSING)



- fast and **reliable** response
- **early warning** nano-enabled detection systems
- **pervasive, interconnected orthogonal** detection system



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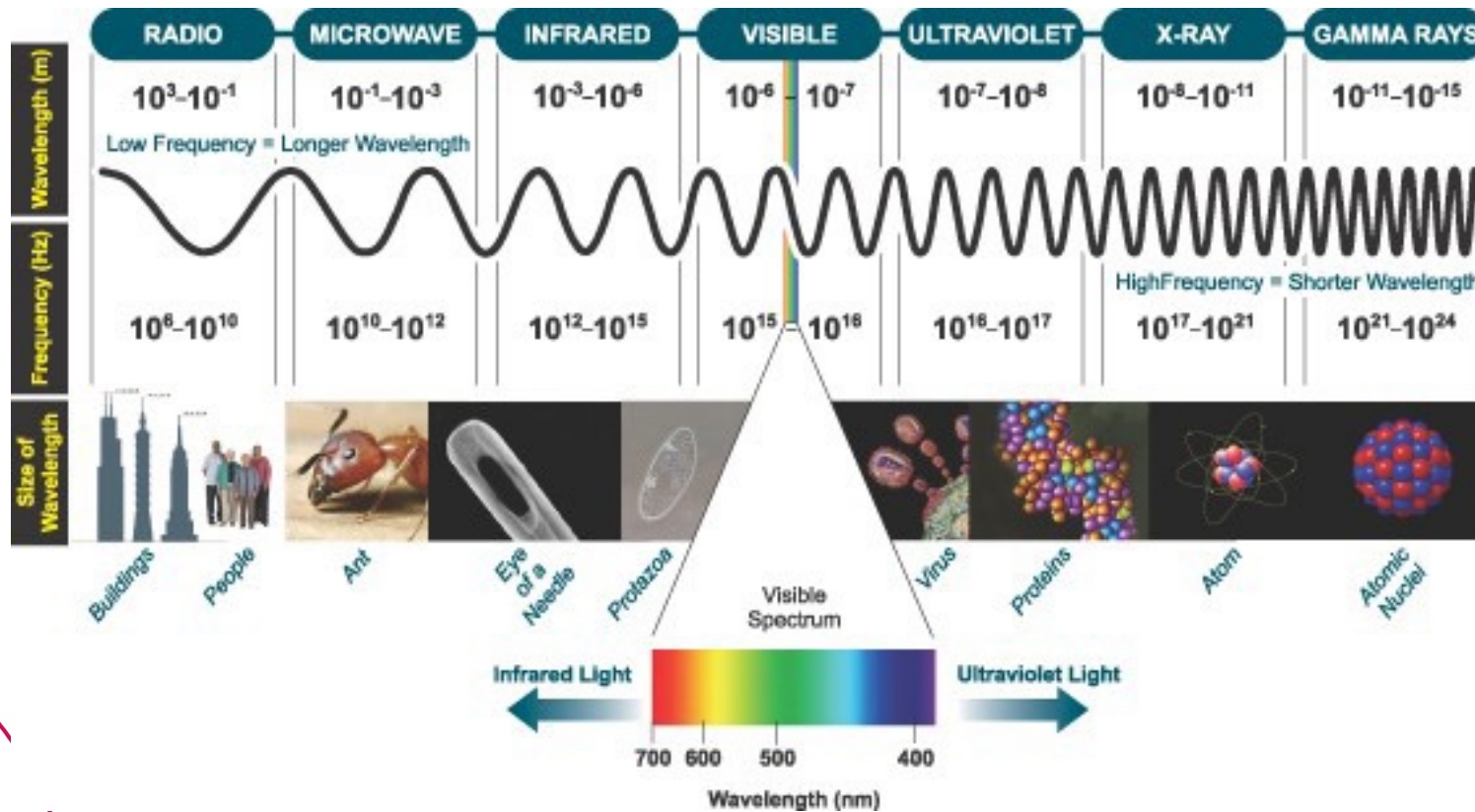
OUTLINE

1. **Chemical Sensing & Scenario Driven Requirements**
2. **SERS Fundamentals**
 - **Spectroscopy & Raman Spectroscopy**
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 - **Measuring SERS**
 - **Portable Instrumentation**
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 - **Challenges & Strategies**
 - **Our Approach**
 - **Main Results**
4. **Take-Home Messages**

2. SERS Fundamentals

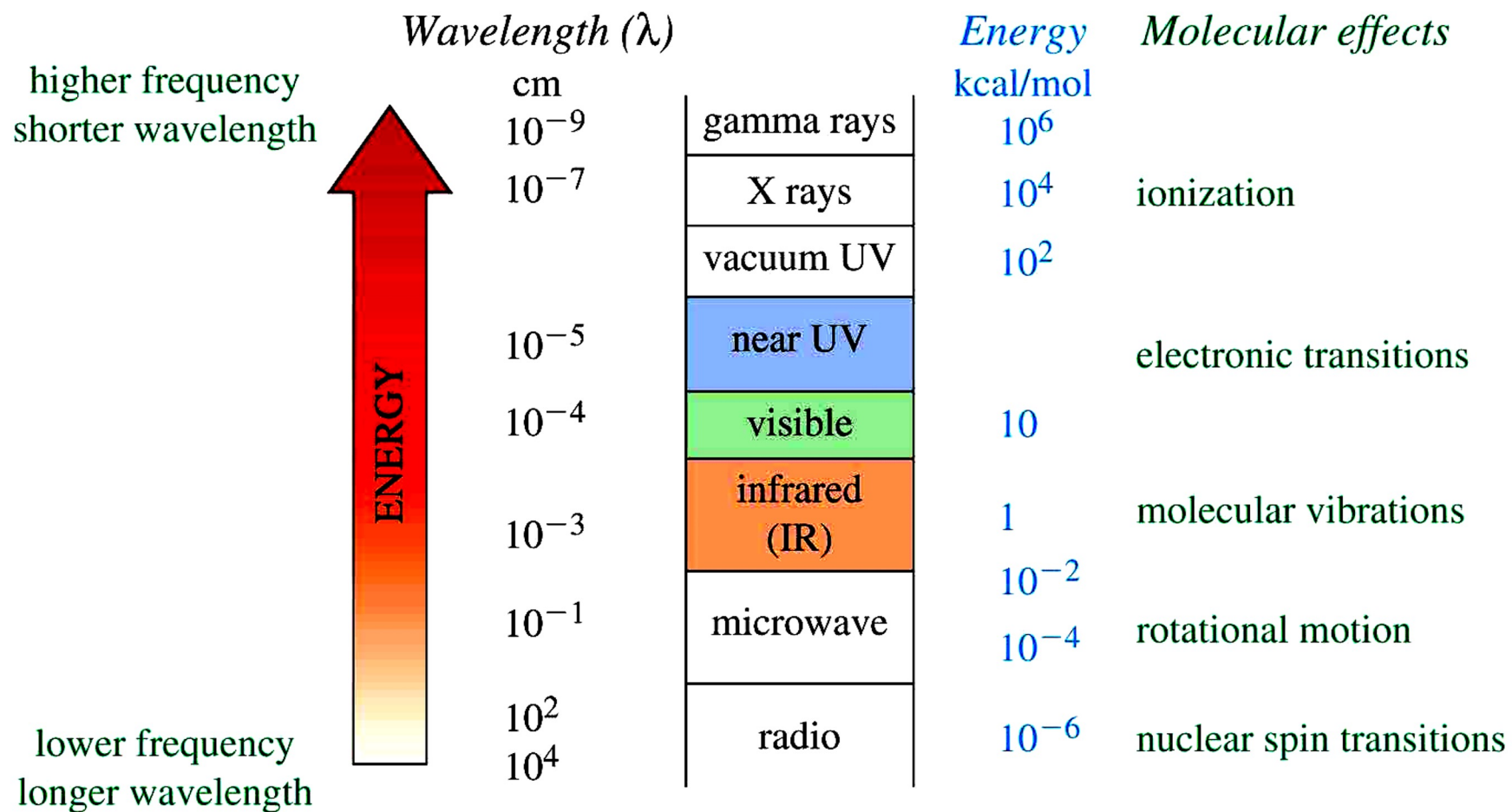
SPECTROSCOPY DEFINITION

It is the branch of science that deals with the study of interaction of electromagnetic radiation with matter.



2. SERS Fundamentals

EFFECT OF ELECTROMAGNETIC RADIATION ON MOLECULES

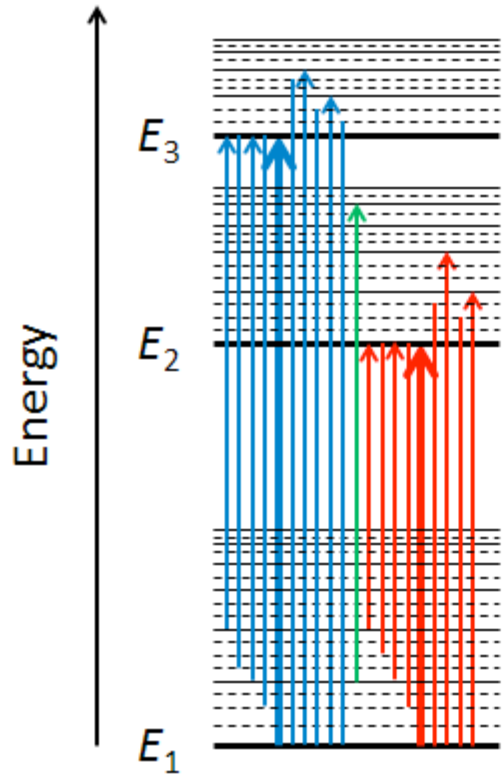


2. SERS Fundamentals

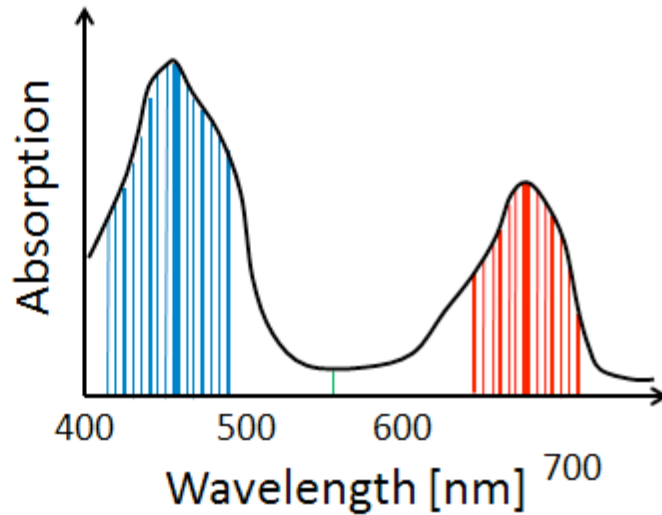
SPECTROSCOPY

Spectrum = the intensity (or flux) of radiation as a function of wavelength

Energy levels



Electronic + vibrational + rotational energy levels



ABSORPTION Spectroscopy

UV-Vis

Electronic levels

IR

Vibrational and rotational levels

EMISSION spectroscopy

Fluorescence

Singlet excited state

Well Stablished Sensing Technologies

Gas Adsorption and Reaction at the MOS surface



Electrochems

Electrochemical Reaction at the Electrode



Metal Oxides



NDIRs

Non Dispersive IR Absorption of Gas Molecules



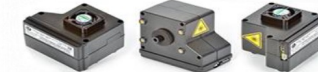
Pellistors (for combustibles)

Pt modified wire to promote gas oxidation



PIDs (for VOCs)

Photolionization and Detection of Gas Molecules



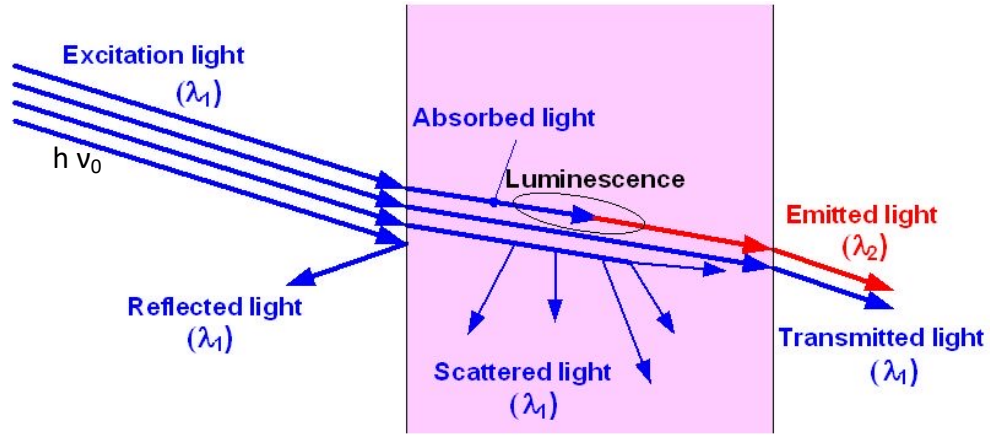
OPCs (for aerosols)

Optical Particle Counters

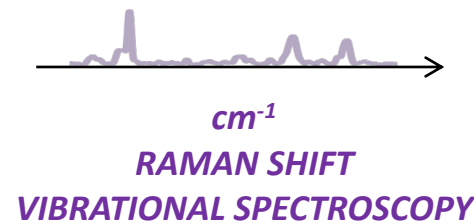
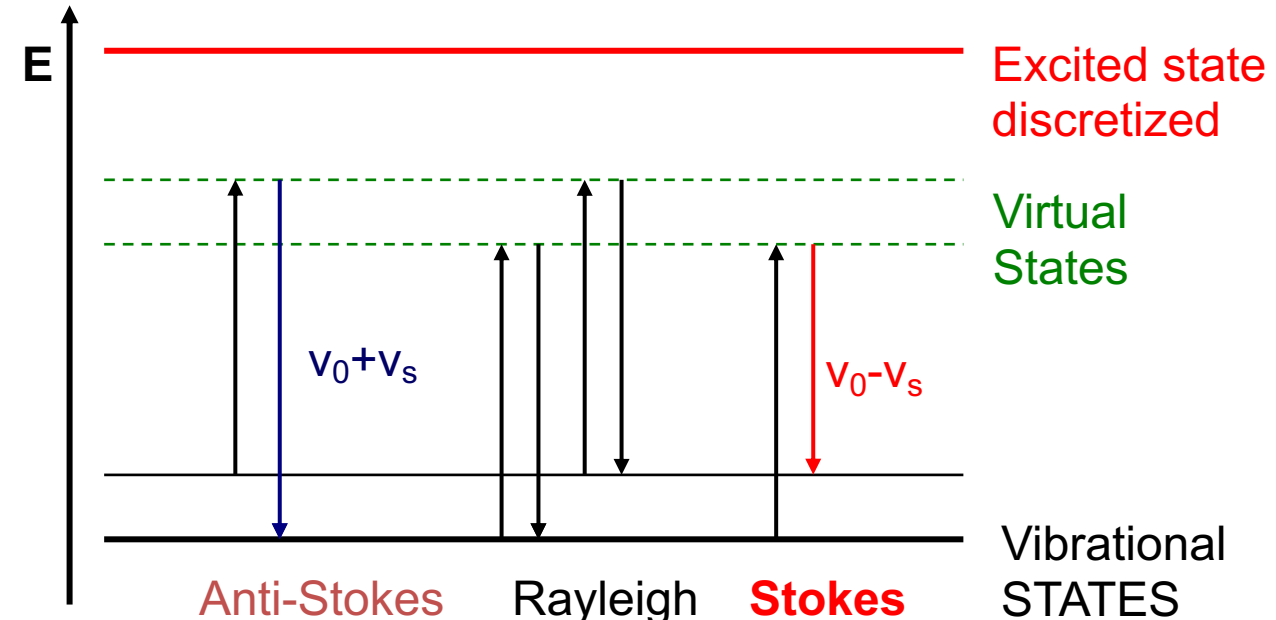
$$\Delta E_{\text{rotational}} < \Delta E_{\text{vibrational}} < \Delta E_{\text{electronic}}$$

2. SERS Fundamentals

RAMAN SPECTROSCOPY: INELASTIC SCATTERING PROCESS (1929's)



- Absorption ν_0 & Molecule/Atoms Polarization to a virtual state
- Elastic Scattering RAYLEIGH (only polarization of electrons): Re-emission of photons at $= \nu_0$
- Inelastic Scattering RAMAN (+ nuclear motion): Re-emission of photons at $\nu_s \neq \nu_0$



2. SERS Fundamentals

RAMAN SCATTERING: INHERENT WEAK EFFECT, POOR RAMAN SIGNALS

1 in every 10^6 – 10^8 photons which scatter is Raman scattered

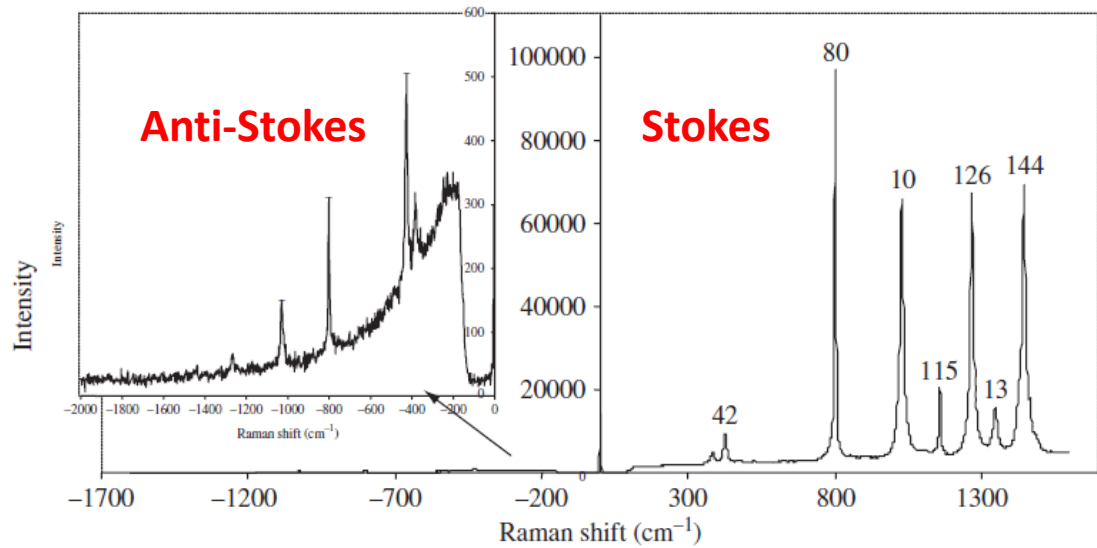
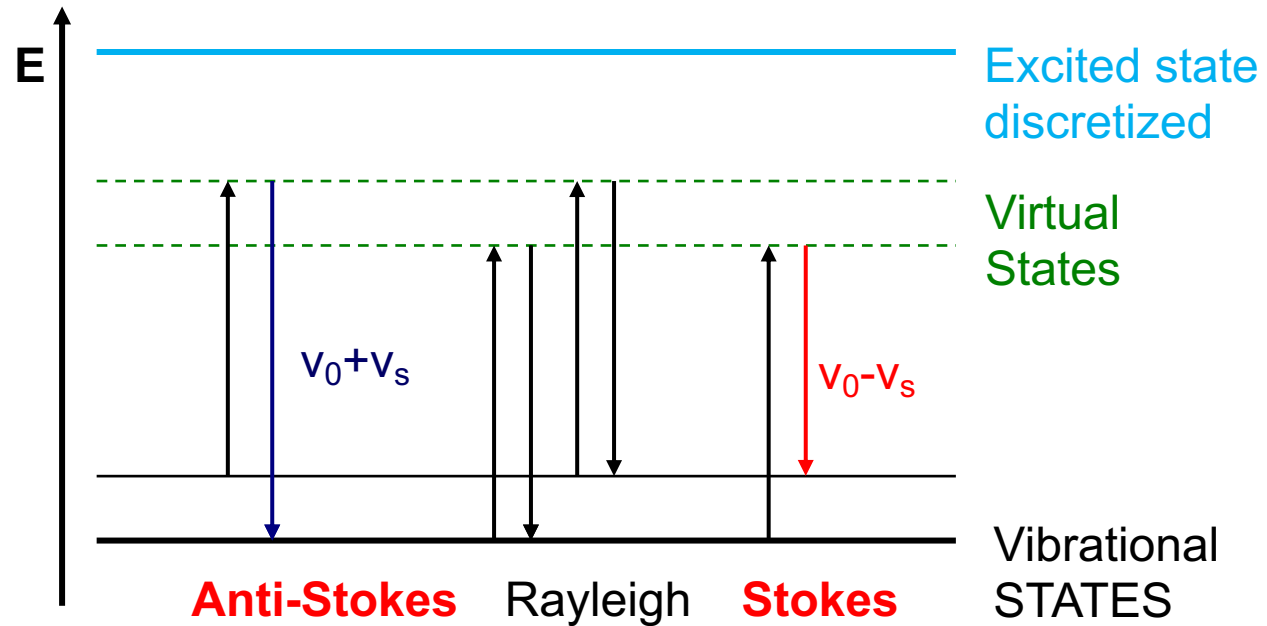


Figure 1.3. Stokes and anti-Stokes scattering for cyclohexane. To show the weak anti-Stokes spectrum, the y-axis has been extended in the inset.

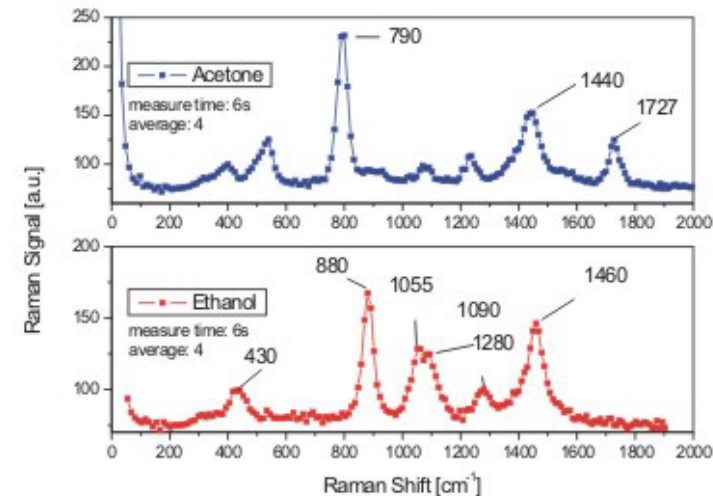
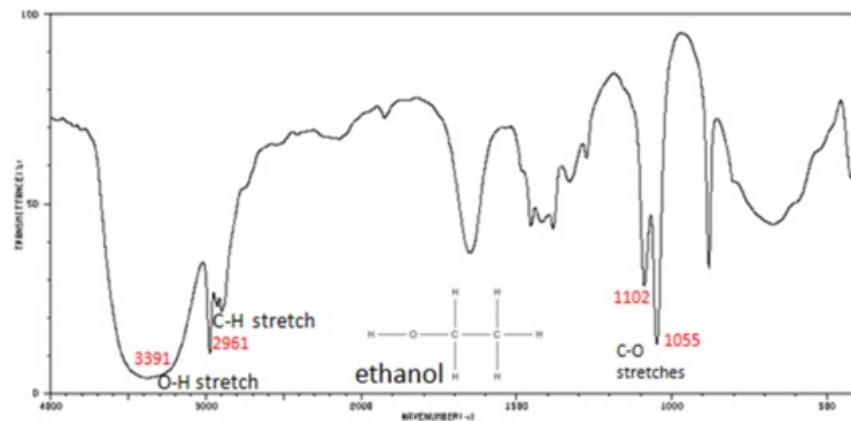


2. SERS Fundamentals

VIBRATIONAL SPECTROSCOPIES

“The main **spectroscopies** to detect “**vibrations in molecules**” are based on the processes of **Infrared Absorption** and **Raman Scattering**”

“They are widely used to provide information on chemical structures and physical forms, to identify substances from the characteristic spectral patterns (**‘fingerprinting’**), and to determine quantitatively or semi-quantitatively the amount of a substance in a sample”



2. SERS Fundamentals

VIBRATIONAL SPECTROSCOPIES: SELECTION RULES

“Intense **Raman** vibrations occurs from vibrations* which causes a **change in the polarizability (distorsion) of the electron cloud** around the molecule”

Bulky molecules, unpaired delocalized electrons: LARGE RAMAN CROSS SECTION

“The most intense **Infrared absorption** occurs by a change in the **dipole moment** of the molecule when the molecule is vibrating”

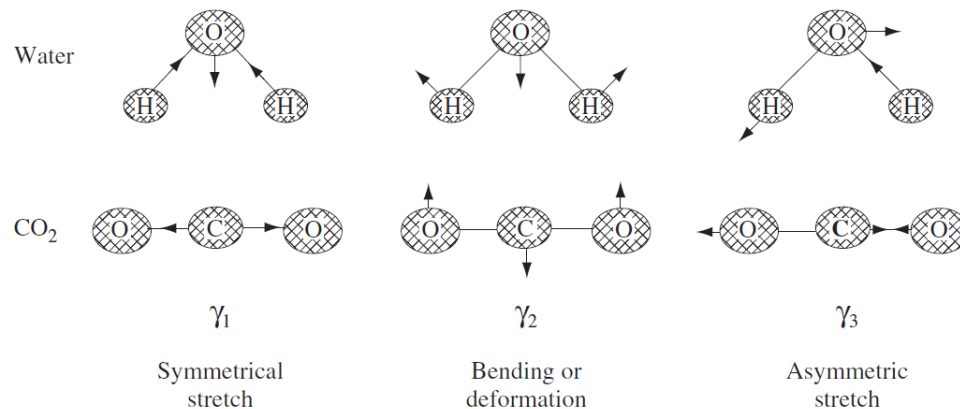


Figure 1.5. Spring and ball model – three modes of vibration for H₂O and CO₂.

- H₂O is poorly Raman scattered
- H₂O is a stronger IR absorber

*Vibrational degrees of freedom in a molecule: $3N-6$ or $3N-5$ (lineal molecules) N : n^o. atoms

| Functional Group/ Vibration | Region | Raman | InfraRed |
|--|------------------------------|--------|----------|
| Lattice vibrations in crystals, LA modes | 10 - 200 cm^{-1} | strong | strong |
| $\delta(\text{CC})$ aliphatic chains | 250 - 400 cm^{-1} | strong | weak |
| $\nu(\text{Se-Se})$ | 290 -330 cm^{-1} | strong | weak |
| $\nu(\text{S-S})$ | 430 -550 cm^{-1} | strong | weak |
| $\nu(\text{Si-O-Si})$ | 450 -550 cm^{-1} | strong | weak |
| $\nu(\text{Xmetal-O})$ | 150-450 cm^{-1} | strong | med-weak |
| $\nu(\text{C-I})$ | 480 - 660 cm^{-1} | strong | strong |
| $\nu(\text{C-Br})$ | 500 - 700 cm^{-1} | strong | strong |
| $\nu(\text{C-Cl})$ | 550 - 800 cm^{-1} | strong | strong |
| $\nu(\text{C-S})$ aliphatic | 630 - 790 cm^{-1} | strong | medium |
| $\nu(\text{C-S})$ aromatic | 1080 - 1100 cm^{-1} | strong | medium |
| $\nu(\text{O-O})$ | 845 -900 cm^{-1} | strong | weak |
| $\nu(\text{C-O-C})$ | 800 -970 cm^{-1} | medium | weak |
| $\nu(\text{C-O-C})$ asym | 1060 - 1150 cm^{-1} | weak | strong |

Continue...

| | | | |
|--|------------------------------|---------------|--------|
| $\nu(\text{CC})$ alicyclic, aliphatic chain vibrations | 600 - 1300 cm^{-1} | medium | Medium |
| $\nu(\text{C}=\text{S})$ | 1000 - 1250 cm^{-1} | strong | weak |
| $\nu(\text{CC})$ aromatic ring chain vibrations | *1580, 1600 cm^{-1} | strong | medium |
| | *1450, 1500 cm^{-1} | medium | medium |
| | *1000 cm^{-1} | strong/medium | weak |
| $\delta(\text{CH}_3)$ | 1380 cm^{-1} | medium | strong |
| $\delta(\text{CH}_2)$ $\delta(\text{CH}_3)$ asym | 1400 - 1470 cm^{-1} | medium | medium |
| $\delta(\text{CH}_2)$ $\delta(\text{CH}_3)$ asym | 1400 - 1470 cm^{-1} | medium | medium |
| $\nu(\text{C}-(\text{NO}_2))$ | 1340 - 1380 cm^{-1} | strong | medium |
| $\nu(\text{C}-(\text{NO}_2))$ asym | 1530 - 1590 cm^{-1} | medium | strong |
| $\nu(\text{N}=\text{N})$ aromatic | 1410 - 1440 cm^{-1} | medium | - |
| $\nu(\text{N}=\text{N})$ aliphatic | 1550 - 1580 cm^{-1} | medium | - |
| $\delta(\text{H}_2\text{O})$ | ~1640 cm^{-1} | weak broad | strong |
| $\nu(\text{C}=\text{N})$ | 1610 - 1680 cm^{-1} | strong | medium |
| $\nu(\text{C}=\text{C})$ | 1500 - 1900 cm^{-1} | strong | weak |
| $\nu(\text{C}=\text{O})$ | 1680 - 1820 cm^{-1} | medium | strong |
| $\nu(\text{C}\equiv\text{C})$ | 2100 - 2250 cm^{-1} | strong | weak |
| $\nu(\text{C}\equiv\text{N})$ | 2220 - 2255 cm^{-1} | medium | strong |
| $\nu(\text{-S-H})$ | 2550 - 2600 cm^{-1} | strong | weak |
| $\nu(\text{C-H})$ | 2800 - 3000 cm^{-1} | strong | strong |
| $\nu(=\text{C-H})$ | 3000 - 3100 cm^{-1} | strong | medium |
| $\nu(\equiv\text{C-H})$ | 3300 cm^{-1} | weak | strong |
| $\nu(\text{N-H})$ | 3300 - 3500 cm^{-1} | medium | medium |
| $\nu(\text{O-H})$ | 3100 - 3650 cm^{-1} | weak | strong |

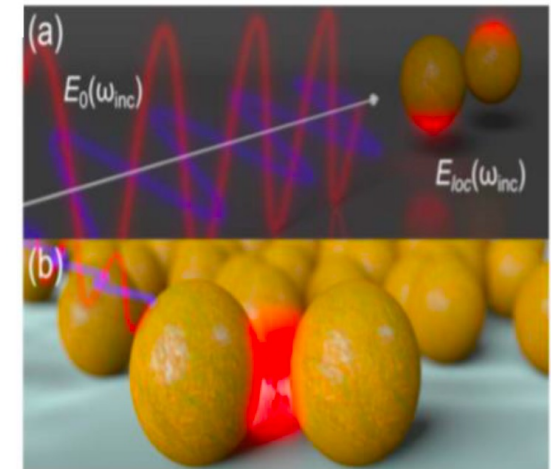
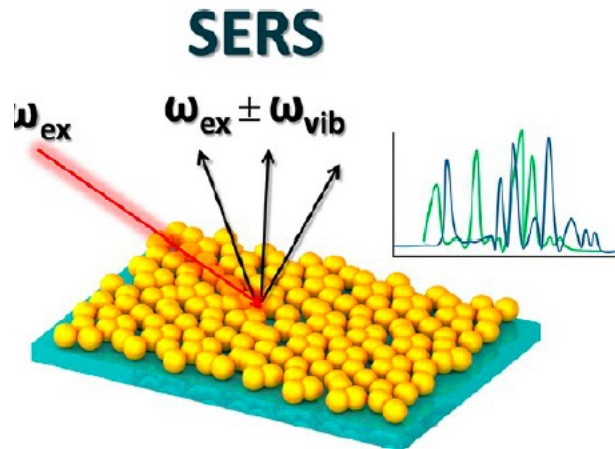
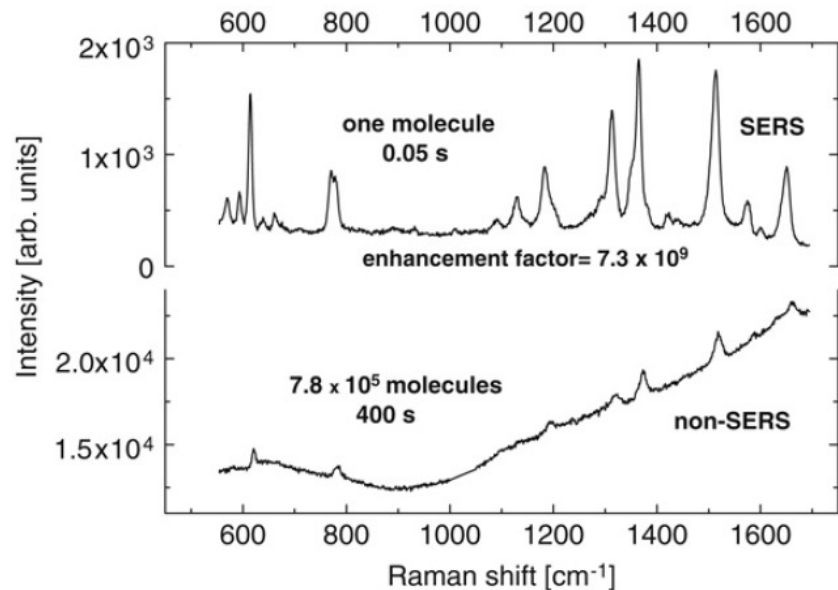
Look for the specific vibrations (literature) before going to the lab/selecting the optical sensor....

2. SERS Fundamentals

Surface Enhanced Raman Spectroscopy (1974's):

...About **amplifying** Raman signals by several orders of magnitude, through the electromagnetic interaction of light with **metals**

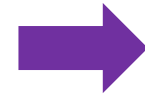
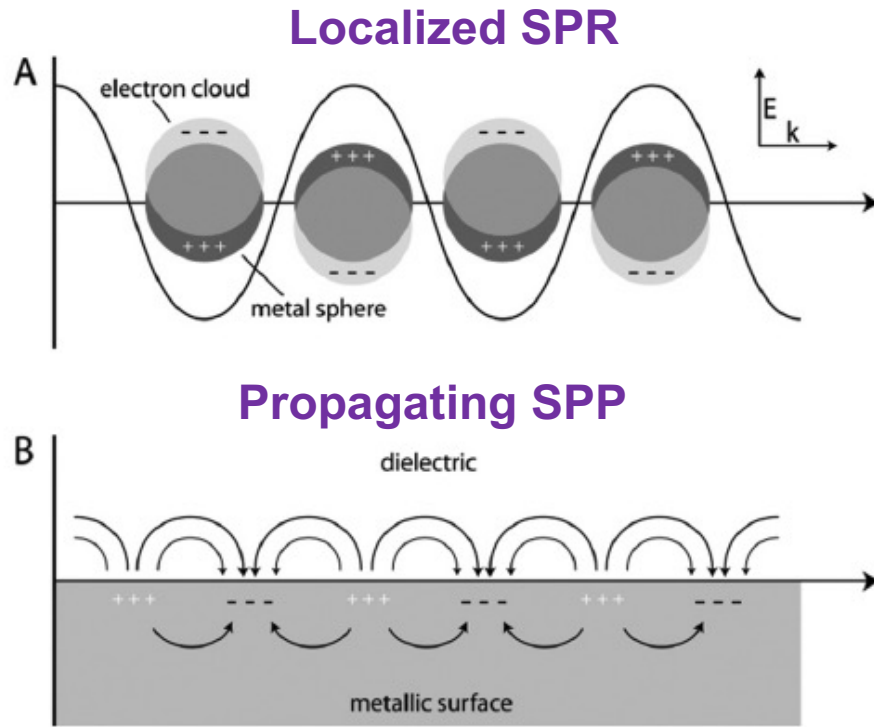
...To profit from these, the molecules must typically be adsorbed on the metal **surface**, or at least very close to it (typ. maximum **10nm**)...



2. SERS Fundamentals

Surface Enhanced Raman Spectroscopy: SERS Effect & substrates

Electromagnetic Enhancement \Rightarrow Metallic Nanostructures \Rightarrow Surface Plasmons



Metals: dielectric function

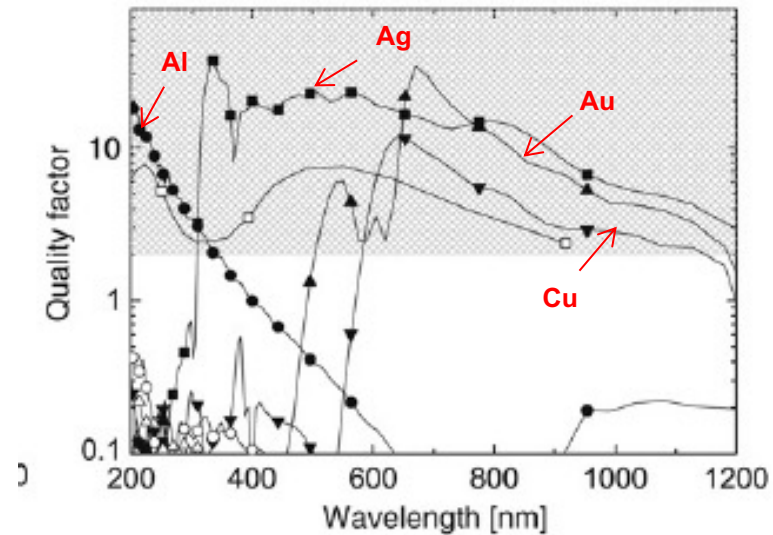


Fig. 1 Schematic illustration of (A) localized and (B) propagating surface plasmon polaritons. E depicts the electric field vector and k the wave vector.



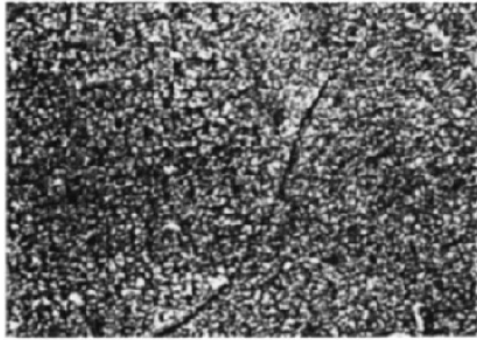
To consider: reactivity of the Surface: Ag vs. Au

- EM enhancement up to 10^8
- Chemical Enhancement up to 10^6

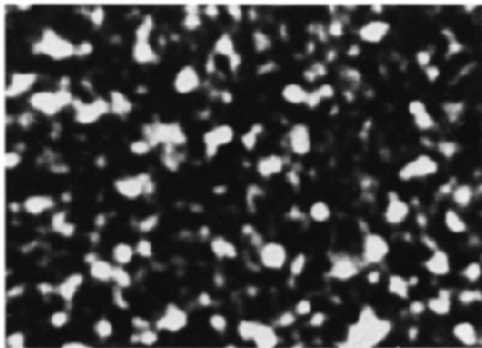
2. SERS Fundamentals

Surface Enhanced Raman Spectroscopy: SERS substrates

Silver coated electrodes (70's)

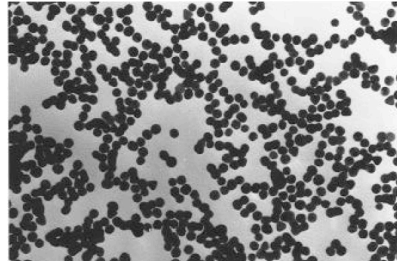


c 1 μm



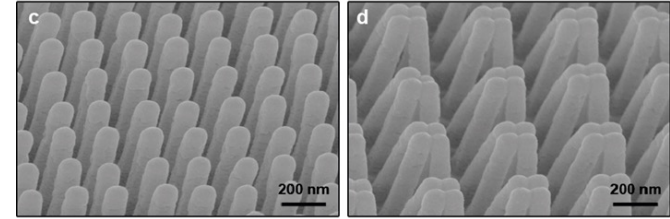
c' 300 nm

Metallic nanoparticles (end 90's)

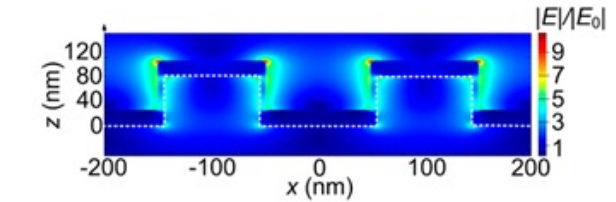
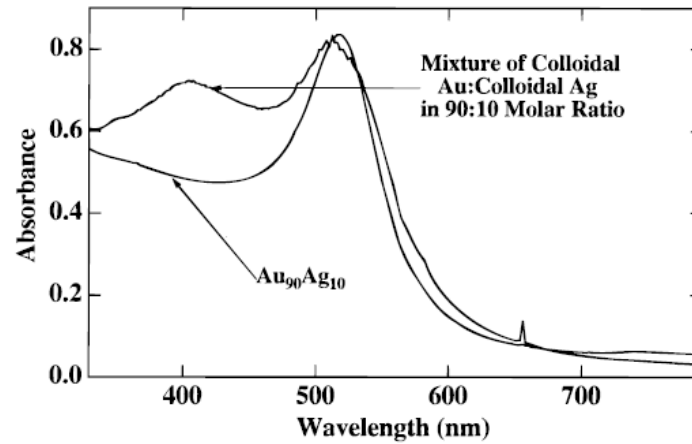


Ordered nanostructures ... last years

Nanophotonics



Hu, M. et al. JACS, 2010, 132, p. 12820-2822.

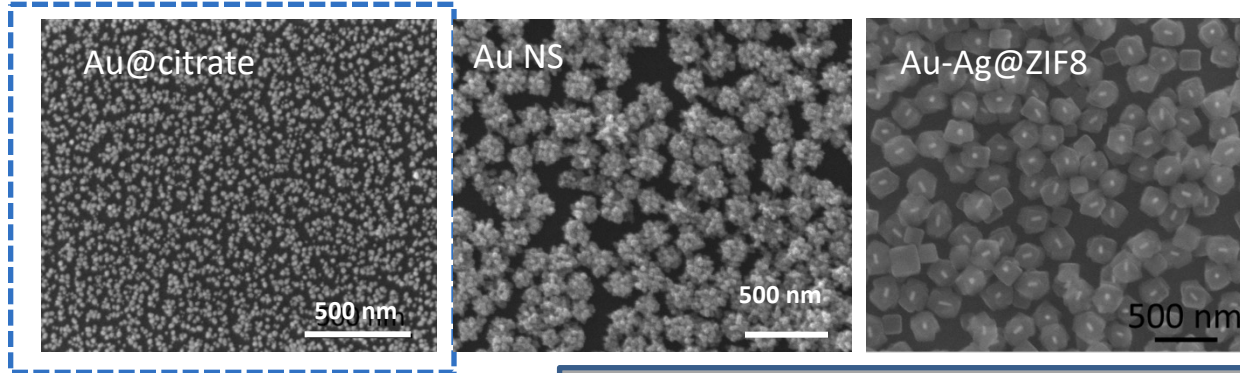


Liu, X. et al. ACS Nano, 2015, 9 (1), p. 260-270.

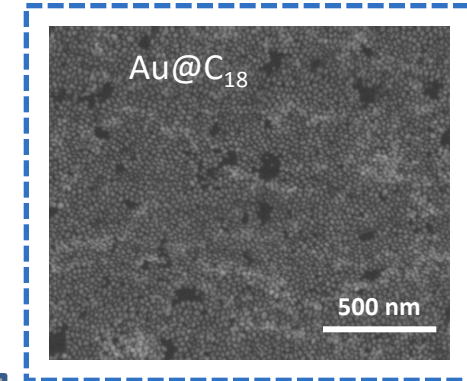
2. SERS Fundamentals

SERS substrates @ INMA

Layer by Layer

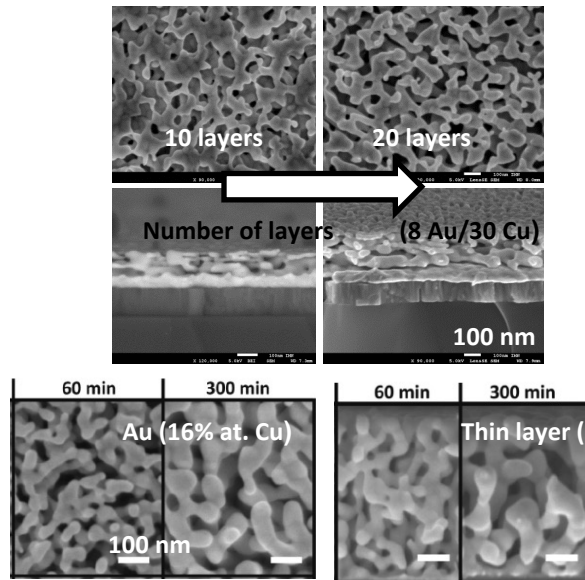


Langmuir-Schaefer

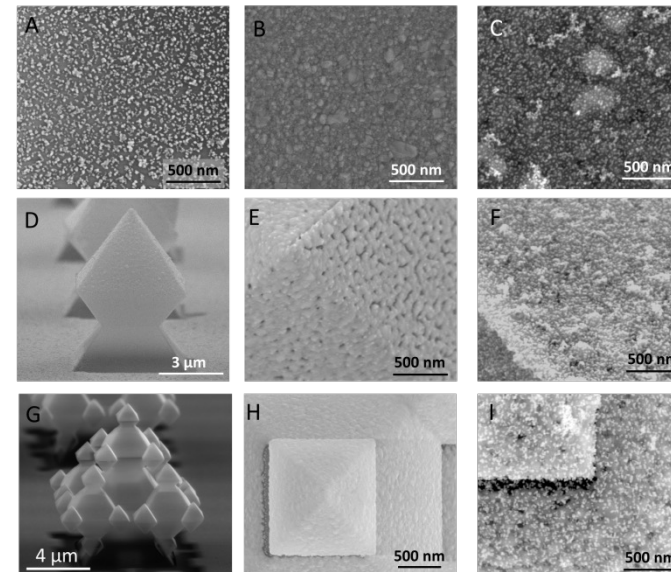


2D SERS substrates “bottom-up”
vs.
3D SERS substrates “top-down”

Institut des Matériaux Jean Rouxel



MESA+ Institute for Nanotechnology



2. SERS Fundamentals

Surface Enhanced Raman Spectroscopy: SERS substrates based on Colloids Transfer

Metallic Nanostructures: Size – Shape - Gap

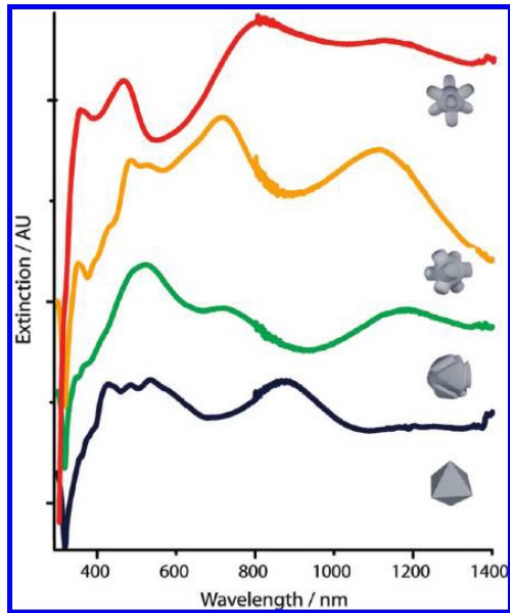


Figure 4. UV-vis-NIR characterization of the octahedral-shaped particles as a function of the etching progress when starting from octahedral nanoparticles. Interestingly, as the etching progresses the octapod structures have LSPs with greater intensity in the red and near-infrared regions.

J. AM. CHEM. SOC. 9 VOL. 132, NO. 1, 2010

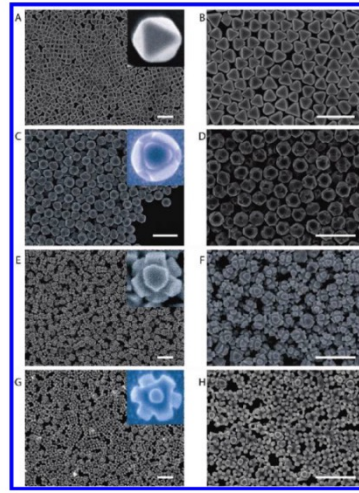


Figure 2. SEM images following the etching progress of the silver octahedral-shaped nanoparticles. (A, B) Octahedra-shaped starting material showing regular size and shape, which are essential for controlled etching reactions; (C, D) using a small amount of etchant, the edges and corners can be selectively etched leaving gaps of 5–10 nm; (E, F) when exposed to a slightly higher concentration of etching solution, eight distinct arms develop; and (G, H) finally, at relatively high concentrations of etching solution, octapod-shaped nanoparticles are isolated in high yield. All scale bars shown represent 1 μm .

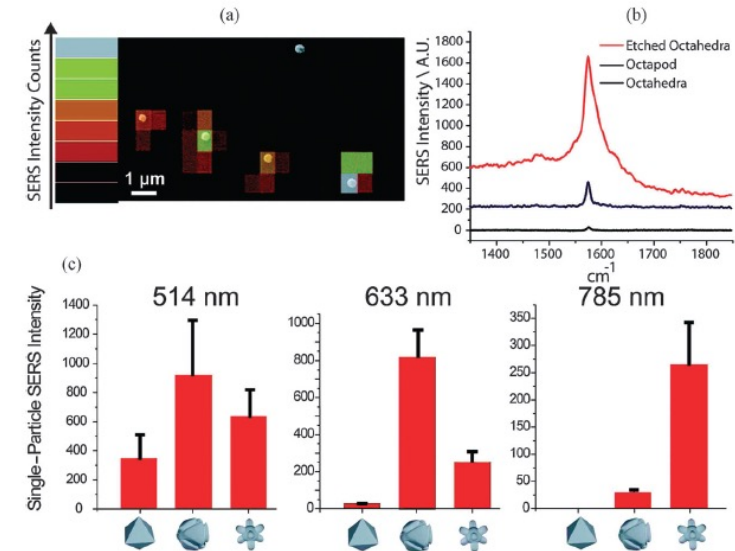
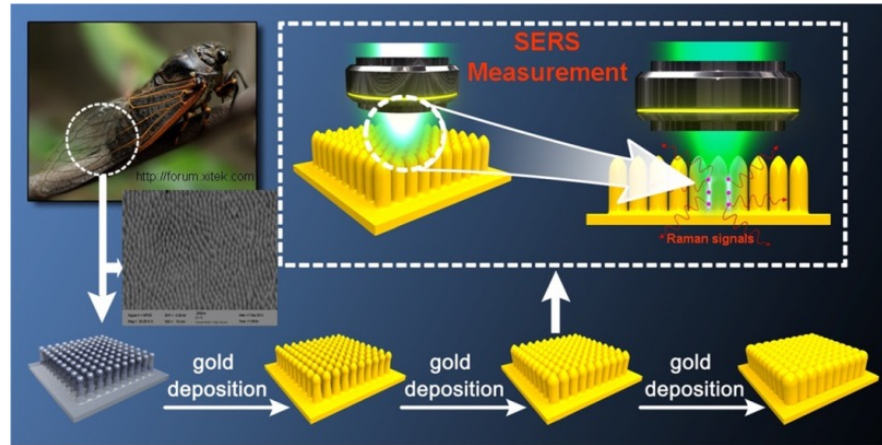


Fig. 8 Surface-enhanced Raman scattering of benzenethiol on single Ag nanoparticles. (a) SEM image of particles overlaid with SERS intensity map of the 1584 cm^{-1} mode of benzenethiol. (b) SERS spectra for each shape particle obtained with 633 nm excitation. (c) Average single particle SERS intensity as a function of shape and excitation wavelength. Reproduced with permission from ref. 168.

2. SERS Fundamentals

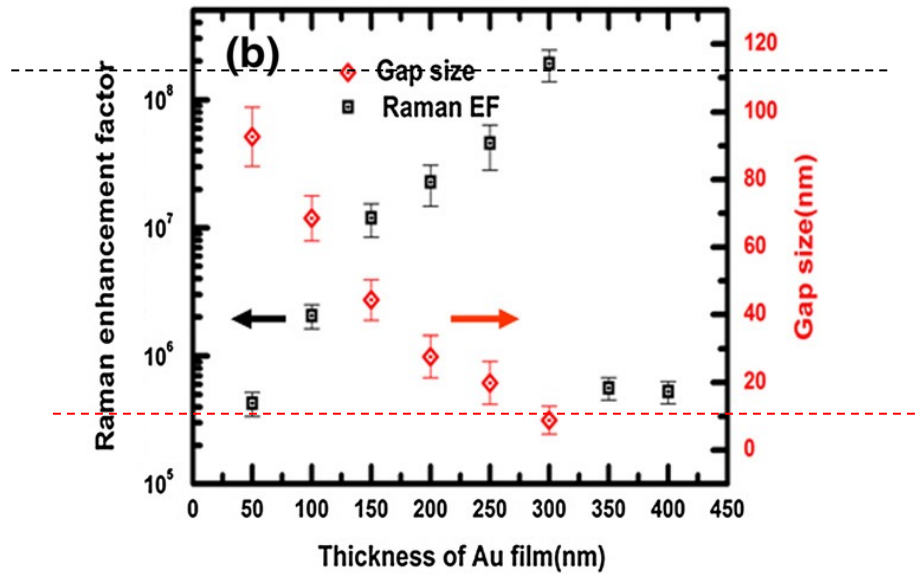
Surface Enhanced Raman Spectroscopy: SERS substrates based on Colloids Transfer

Metallic Nanostructures: Size – Shape - Gap



Nanoscale Research Letters, 2013, 8, 437

HOT SPOTS FORMATION



2. SERS Fundamentals

Measuring SERS

SERS probes (analytes or labels)

- Dyes produce larger signals in Raman and are typical probes but....(Raman intensity is characterized by Raman cross-section)
- Can any molecule be measured with SERS?
- Can any molecule be attached (or at least brought close) to a metallic substrate?
- Will the resulting SERS signal be sufficiently strong to be observed (and distinguished from any other unavoidable signal and noise)?
- Photo stability of the probe (power, time, dose laser line) should be considered



Necessary to optimize acquisition parameters to maximize the enhancement

2. SERS Fundamentals

Measuring SERS: Analytical Enhancement Factor (AEF) or SERS Gain

$$AEF = \frac{I_{SERS}/c_{SERS}}{I_{RS}/c_{RS}}$$

- Measurement is easily implemented and reproducible
- Identical Raman and SERS acquisition conditions
- Shortcomings:
 - C_{SERS} is clearly affected by the adsorption efficiency (intercomparison conditions)
 - C_{RS} carefully selected for dyes to avoid photobleaching, molecular aggregation, molecular absorption... (low concentrations are preferable)

2. SERS Fundamentals

$$AEF = \frac{I_{SERS}/c_{SERS}}{I_{RS}/c_{RS}}$$

Measuring SERS: Analytical Enhancement Factor (AEF)

Table 1 Characteristics of SERS hot spot substrates discussed in the text. Values are a best estimate given reported literature values. See text for details. The following are acronyms used in the table: MFONs—metal film over nanospheres; TERS—tip-enhanced Raman spectroscopy; SHINERS—shell-isolated nanoparticle enhanced Raman spectroscopy; NR—not reported; NA—not applicable; Fs—femtosecond

| | EF range | Estimated gap size (nm) | Cost (\$-\$\$) | Analyte specificity | Proven SM sensitivity | Ref. |
|-------------------------------------|-----------------------|-------------------------|----------------|---------------------|-----------------------|--------|
| Non-hot spot substrates | | | | | | |
| Island films | 10^4 | 10–100 | \$\$ | No | Yes | 22–24 |
| Periodic particle array | 10^7 | 10–100 | \$\$ | No | No | 4, 25 |
| Nanorods | 10^3 – 10^7 | NA | \$ | No | No | 26 |
| Nanoshells | $\sim 10^6$ | NA | \$ | No | No | 27 |
| Fabricated substrates | | | | | | |
| MFONs | 10^3 – 10^{11} | 0.1–10 | \$\$ | No | No | 28, 29 |
| Lace-shell NPs | $\sim 10^2$ | 1–10 | \$\$ | No | No | 30 |
| Fs laser etching | $< 10^9$ | NR | \$\$ | No | No | 31, 32 |
| E-beam lithography | 10^3 – 10^5 | 1–100 | \$\$ | No | No | 33–35 |
| Templated nanowires | 10^2 – 10^3 | 5–100 | \$\$ | No | No | 36–38 |
| Bowtie electromigration | 10^8 | 0.1–10 | \$\$ | No | No | 39 |
| Spheres on E-beam posts | 10^8 | 0–1 | \$\$ | Yes | No | 40 |
| Assembled structures | | | | | | |
| Aggregated colloids | 10^6 – 10^{10} | 0–10 | \$ | No | Yes | 7, 41 |
| DNA hybridization | 10^8 – 10^{12} | 1–100 | \$\$ | Yes | Yes | 42–44 |
| Embedded polymers | $\sim 10^5$ | 10–100 | \$ | No | No | 45, 46 |
| SiO ₂ encapsulated cores | 10^6 – 10^8 | 0–10 | \$\$ | Yes | No | 47 |
| Particles in wells | 10^9 | 0–10 | \$ | No | No | 48, 49 |
| Nanofingers | 10^{10} – 10^{11} | 0–10 | \$\$ | No | No | 50, 51 |
| TERS | 10^4 – 10^9 | 0–100 | \$\$ | No | Yes | 52–54 |
| SHINERS | 10^5 – 10^8 | 1–10 | \$ | No | No | 55, 56 |

Huge variability due to the long list of influencing parameters (apart from the substrate):

- Laser excitation, Detection Set-up
- SERS probe: Raman polarizability, adsorption efficiency, analyte concentration, preferential orientation
- **CHEM mechanism: molecule-substrate specific**

2. SERS Fundamentals

Measuring SERS: AEF vs. Spatial uniformity-Reproducibility

Table 2 Distribution of EFs on AgFON substrate. Adapted with permission from ref. 29

| Raman enhancement factor | Percentage of molecules | Percent contribution to average SERS signal |
|--------------------------------------|-------------------------|---|
| $< 2.8 \times 10^4$ | 0% | 0% |
| 2.8×10^4 to 1×10^5 | 61% | 4% |
| 10^5 to 10^6 | 33% | 11% |
| 10^6 to 10^7 | 5.1% | 16% |
| 10^7 to 10^8 | 0.7% | 22% |
| 10^8 to 10^9 | 0.08% | 23% |
| 10^9 to 10^{10} | 0.006% | 17% |
| $> 10^{10}$ | 0.0003% | 7% |

ii < 1% of the target molecules adsorbed on the hot spots accounts for > 69% SERS signal!!

- Are Hot Spots really beneficial for “real” applications?
- SERS fluctuations; SERS blinking; Photobleaching...

Average EF (Substrate) vs. Single Molecule EF (SMEF)

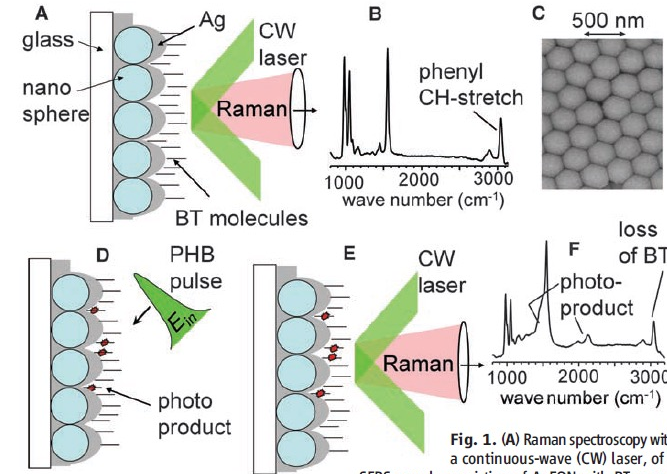


Fig. 1. (A) Raman spectroscopy with a continuous-wave (CW) laser, of a SERS sample consisting of AgFON with BT monolayer. (B) SERS spectrum of BT. (C) Scanning electron micrograph of AgFON surface. (D) The sample was exposed to an intense PHB pulse with laser field E_{in} . BT molecules at sites with local field enhancement g were damaged if $gE_{in} \geq E_{th}$, where E_{th} is the threshold field needed to damage BT. (E and F) The Raman spectrum after PHB shows loss of BT plus new transitions from photoproduct molecules. The loss of BT is quantified by using the integrated area of the phenyl CH-stretch transition at 3050 cm^{-1} .

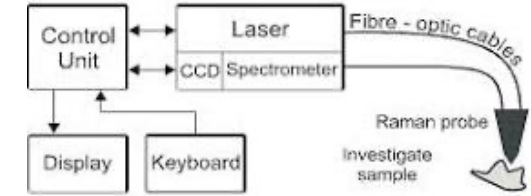
SCIENCE VOL 321 18 JULY 2008



2. SERS Fundamentals

Measuring SERS: Hand-held Raman spectrometers

Key advantage for CBRNe applications: liquid/solid phase material identification through container walls (optically transparent and even semi-opaque)



Handheld systems

| | | | | | | | | | |
|--------------------|---|---|---|--|---|--------------------------------|--|----------------------------------|----------------------------------|
| | | | | | | | | | |
| Spec | B&W Tek NanoRam [®] | B&W Tek TacticID [®] -GP | B&W Tek TacticID [®] -N | Bruker Bravo | Cobalt Resolve | ThermoFisher Scientific Gemini | ThermoFisher Scientific FirstDefender RMX, FirstDefender RM, TruNarc, TruScan GP, TruScan RM | Rigaku Progeny RESQ | Rigaku Progeny |
| Lasers | 785nm | 785nm | 785nm | 700 - 1100 nm (Duo LASER™) 758 nm and 852 nm | 830 nm | 785 nm | 785 nm | 1064 nm | 1064 nm |
| Laser output power | 300mW Max Adjustable in 10% Increments | 300mW Max Software Adjustable | 300mW Max Software Adjustable | low laser power, < 100 mW for both LASERS | Adjustable laser power (475 mW maximum) | 300mW | 300mW | 30-495 mW Adjustable laser power | 30-495 mW Adjustable laser power |
| Range | 176cm ⁻¹ to 2900cm ⁻¹ | 176cm ⁻¹ to 2900cm ⁻¹ | 176cm ⁻¹ to 2900cm ⁻¹ | 300cm ⁻¹ to 3200cm ⁻¹ | | 250-2875cm ⁻¹ | 250-2875cm ⁻¹ | 200-2500cm ⁻¹ | 200-2500cm ⁻¹ |
| Resolution | ~ 9cm ⁻¹ @ 912nm | ~ 9cm ⁻¹ @ 912nm | ~ 9cm ⁻¹ @ 912nm | 10-12 cm ⁻¹ | | 7-10.5 cm ⁻¹ | 7-10.5 cm ⁻¹ | 8-11 cm ⁻¹ | 8-11 cm ⁻¹ |
| Detector | TE-Cooled Linear CCD Array | Linear CCD Array | Linear CCD Array | CCD | | | | TE Cooled InGaAs | TE Cooled InGaAs |
| Other | Point-and-shoot | Point-and-shoot | Point-and-shoot | Point-and-shoot | Point-and-shoot SORS™ | FT-IR combined | Point-and-shoot | Point-and-shoot | Point-and-shoot |

2. SERS Fundamentals

Measuring SERS: Hand-held Raman spectrometers

Key advantage for CBRNe applications: liquid/solid phase material identification through container walls (optically transparent and even semi-opaque)

| | | | | | | | | |
|--------------------|-------------------------|---|---|---|---------------------------|--------------------------------|--|---|
| | | | | | | | | |
| Spec | SersTech SersTech100 | Metrohm Mira M-1 M-2 M-3 | Snowy Range Instruments SORI CBEx | Ocean Optics iDRaman Mini | SmithsDetection ACE-ID | Airsense Analytics LS-ID | TSI ASSURx Raman Handheld Analyzer | TSI EZRaman-H Series Raman Handheld Analyzer |
| Lasers | 785 nm | 785 nm 1064 nm | 785 nm 1064 nm | 633 nm 785 nm | 785 nm | 785 nm | 785 nm | 785 nm |
| Laser output power | | ≤ 100 mW | ≤ 100 mW | 100 mW | | 100mW | 300mW | 300mW |
| Range | | 400cm ⁻¹ to 2300cm ⁻¹ | 400cm ⁻¹ to 2300cm ⁻¹ | 400-2300 cm ⁻¹ | | | 250-2350cm ⁻¹ | 250-2350cm ⁻¹ |
| Resolution | | 12-14 cm ⁻¹ | 12-14 cm ⁻¹ | 16 cm ⁻¹ (633 nm) 18-20 cm ⁻¹ (785 nm) | | | 6 cm ⁻¹ | 6 cm ⁻¹ |
| Detector | | | | CCD | | | CCD | CCD |
| Other | Point-and-shoot | Point-and-shoot ORS | Point-and-shoot ORS | Point-and-shoot ORS | Point-and-shoot ORS | Point-and-shoot | Point-and-shoot | Point-and-shoot |

Standoff material identification
Metrohm, Mira accessory
Distance: up to 1.5 m



2. SERS Fundamentals

Application Fields

Biomolecule Sensing & Bioimaging



- ...degenerative disorders¹
- ...infectious diseases²
- ...genetic diseases³
- ...cancer⁴



¹Bhowmik et al., *ACS Nano*, 2015.
²Shanmukh et al., *Nano Lett.*, 2006.
³García-Rico et al., *Chem. Soc. Rev.*, 2018.
⁴Pazos et al., *J. Am. Chem. Soc.*, 2016

Environmental Monitoring & Industrial Control



- ...environmental toxins⁵
- ...chemicals⁶
- ...heavy metals⁷
- ...pesticides⁸
- ...monitoring chemical reactions⁹
- ...discrimination quiral molecules¹⁰

⁵Feng et al. *Biosens. Bioelectron.*, 2016.
⁶Li et al. *Nanoscale*, 2016.
⁷Xu et al. *Biosens. Bioelectron.*, 2015.
⁸Lafuente et al. *ACS Appl. Mat. Interfaces*, 2020.
⁹Han et al. *ACS Appl. Mater. Interfaces*, 2017.
¹⁰Tkachenko et al. *Nat. Commun.*, 2014.

Raman Spectroscopy



Nano and Micro-technology

✓ Spectral Fingerprint →
MOLECULAR SELECTIVITY/SPECIFICITY

✗ **Low signal (inelastic scattering)**

✓ Universal Technique

🔍 **Raman Cross Section (polarizability)**

✓ Outperforming Portable Instrumentation

✓ Surface Plasmon Resonance
Engineering → **“SINGLE MOLECULE
DETECTION”**

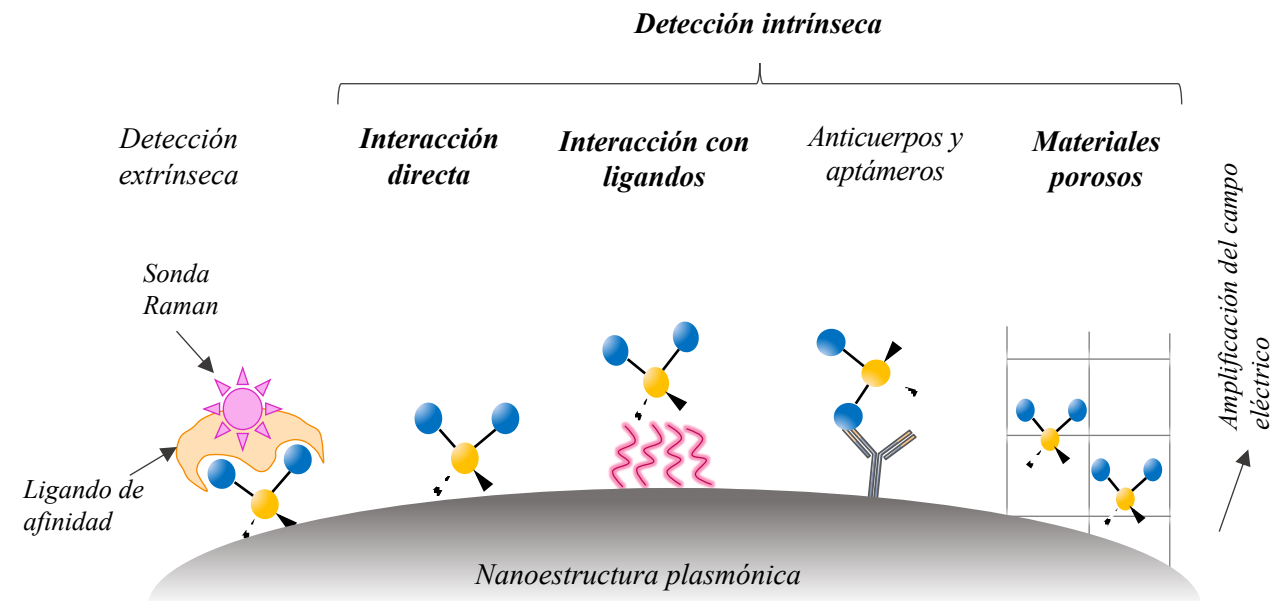
🔍 **SERS effect is a Surface Effect →
DETECTION IN GAS PHASE???**

OUTLINE

1. **Chemical Sensing & Scenario Driven Requirements**
2. **SERS Fundamentals**
 - **Spectroscopy & Raman Spectroscopy**
 - **Surface Enhanced Raman Spectroscopy: SERS Effect & substrates**
 - **Measuring SERS**
 - **Portable Instrumentation**
 - **Application Fields**
3. **SERS for Gas Sensing**
 - **Challenges & Strategies**
 - **Our Approach**
 - **Future Work**
4. **Take-Home Messages**

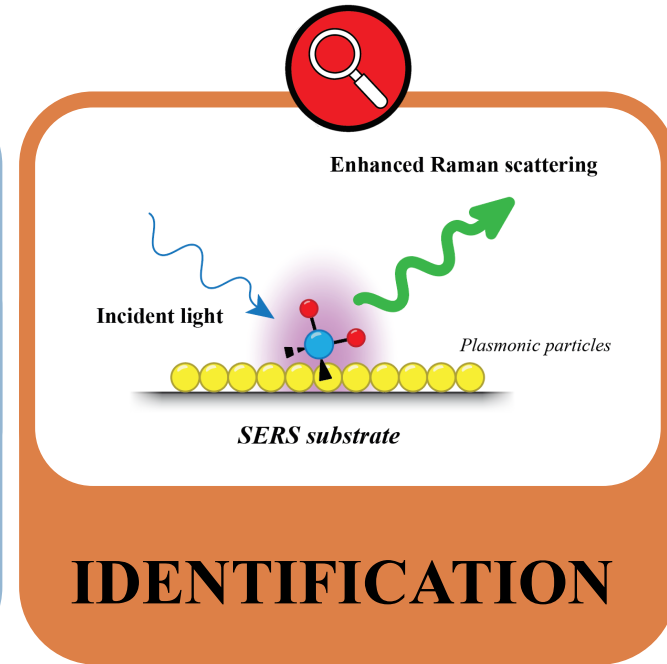
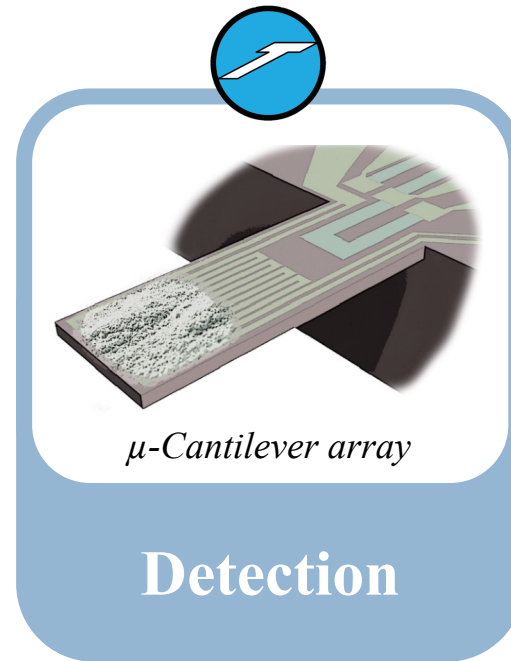
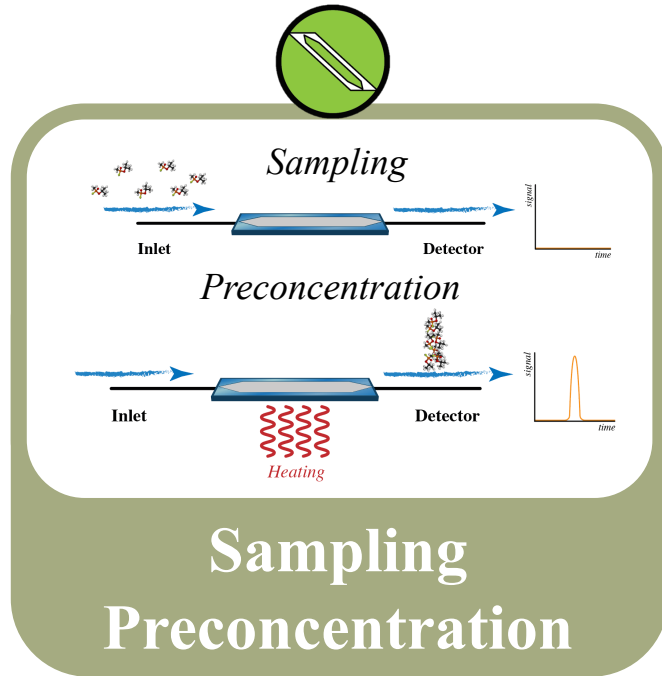
3. SERS for Gas Sensing: Challenges & Opportunities

- Gas phase: low number of molecules per unit volumen $\rho_{\text{gas}} = \rho_{\text{liq}}/1000$
- Low Sticking probability of gas/vapor molecules on the Enhancing Surface: low C_{SERS}
- Raman Cross- Section of Target Molecules
- Engineering SERS susbtrates/chips
 - Electrochemical Assisted SERS
 - Temperature Assisted SERS
 - Surface Chemistry
 - Partition Layers or Porous Surfaces



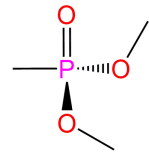
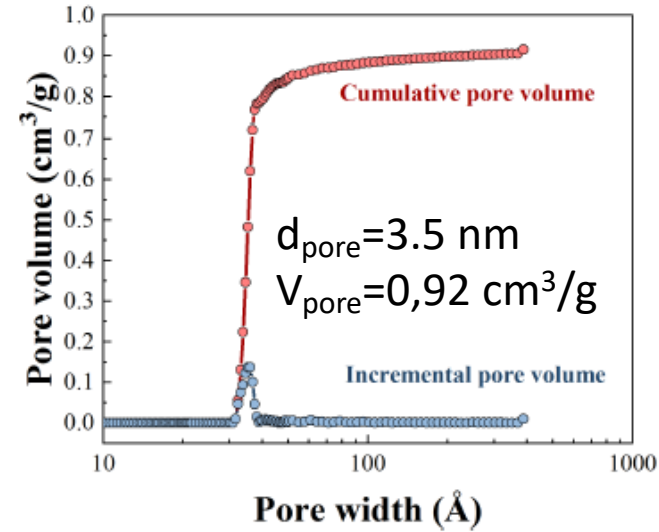
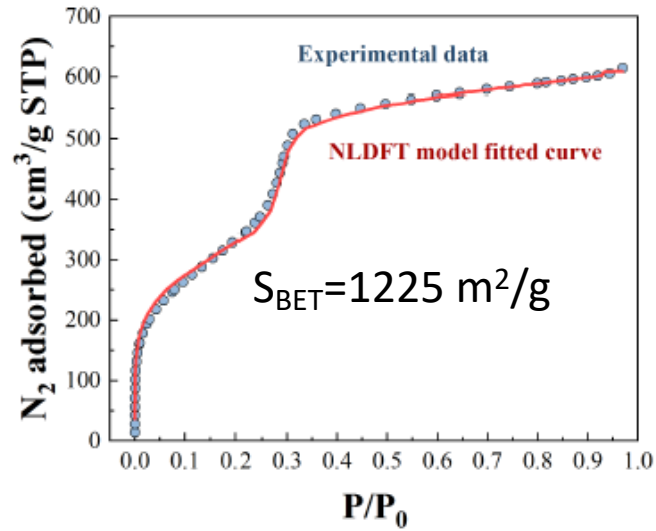
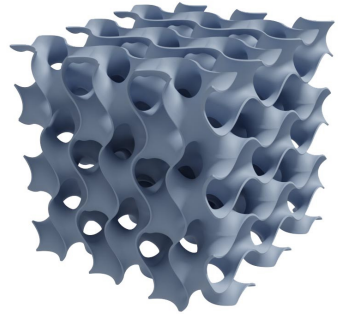
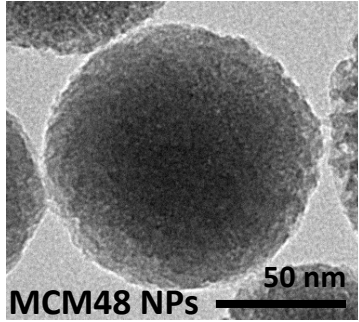
3. SERS for Gas Sensing

Our Approach: From Nanoporous Sorbents to Plasmonic Sorbents



3. SERS for Gas Sensing

Our Approach: From Nanoporous Sorbents to Plasmonic Sorbents



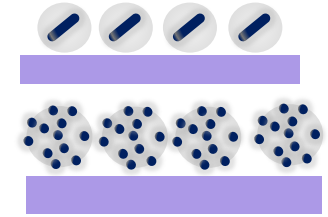
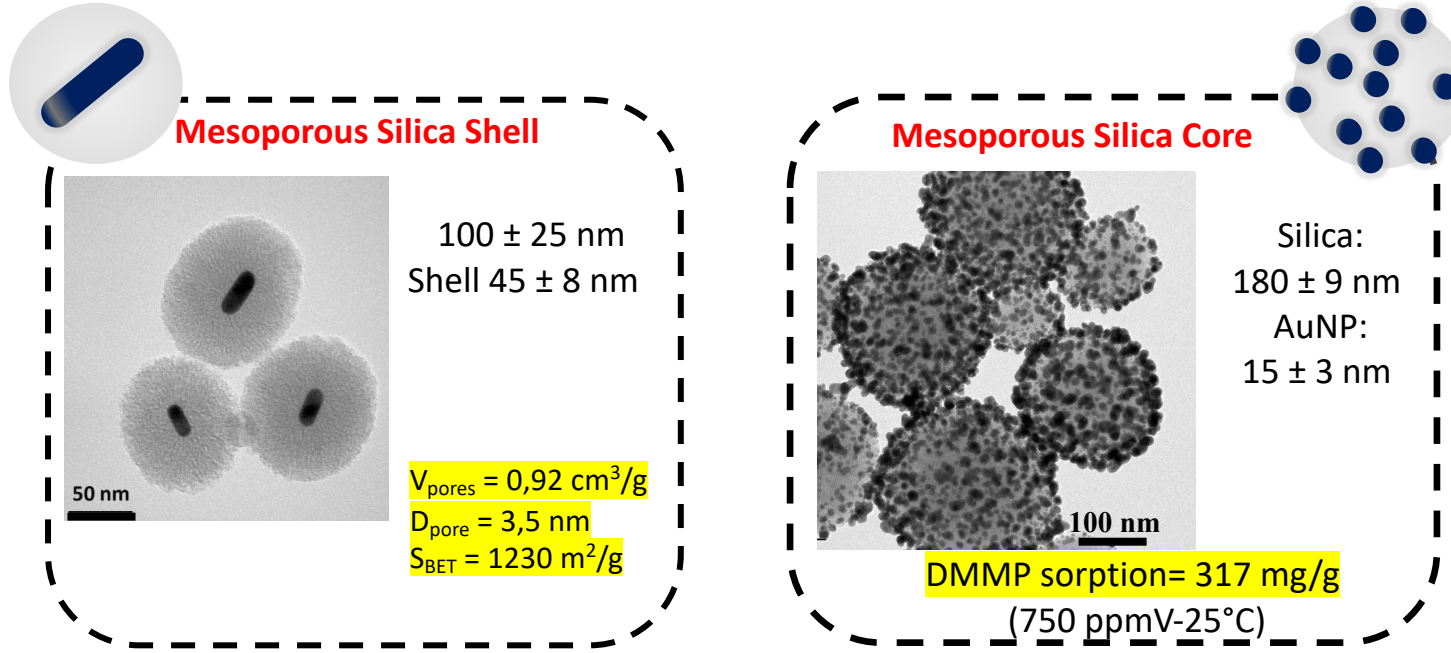
DMMP
Sarin Gas Surrogate

| Material | Control | | After exposure to 747 ppmV DMMP | | | | | |
|------------------------|---------------------------------------|-----------|---------------------------------------|-----------|---------------------------------------|-----------|---------------------------------------|-----------|
| | Δm mg/g _{sorbent} | T (°C) | Δm mg/g _{sorbent} | T (°C) | Δm mg/g _{sorbent} | T (°C) | Δm mg/g _{sorbent} | T (°C) |
| Dense SiO ₂ | 66.8 | 36 | 21.5 | 52 | - | - | - | - |
| MCM48 | 31.4 | 42 | 40.0 | 39 | 132.8 | 97 | 222.7 | 170 |
| MCM48-NH ₂ | 62.4 | 45 | 109.5 | 39 | 82.1 | 61 | 51.6 | 110 |



3. SERS for Gas Sensing: Our Approach

❖ Films based on Core-Shell nanoparticles based on Mesoporous Silica

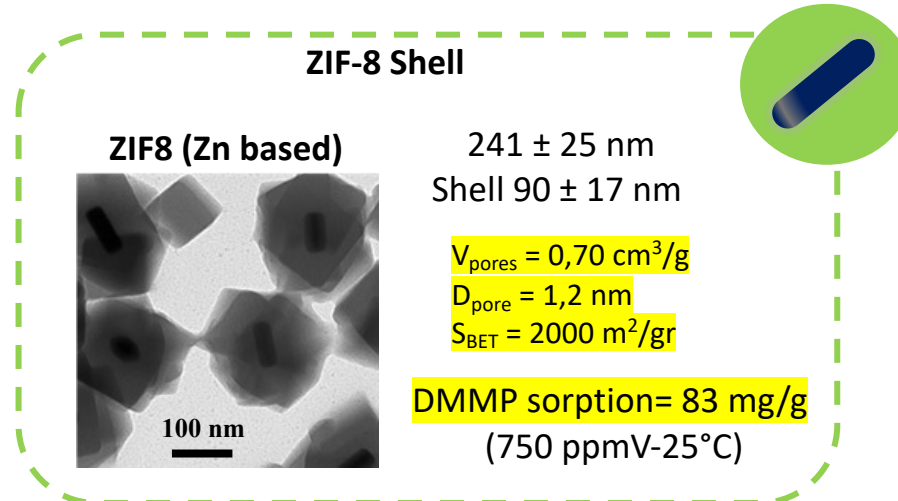


❖ **Ordered Mesopores 2.8 to 3.5 nm in pore size**

❖ **User-friendly Functionalization of Mesoporous Silica**

❖ **Higher DMMP Uptake on Mesoporous Silica**

❖ **Films based on Au@Ag nanorods encapsulated in Microporous ZIF8 (Zn based MOF)***

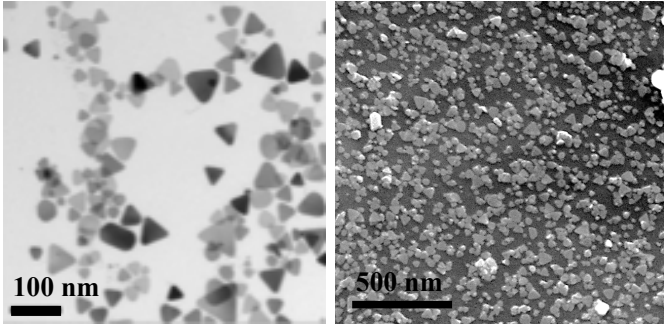


❖ **Thickness of the Porous Moities: preconcentration effect vs response time & LSPR position**

*<https://doi.org/10.1021/acssensors.1c00178>

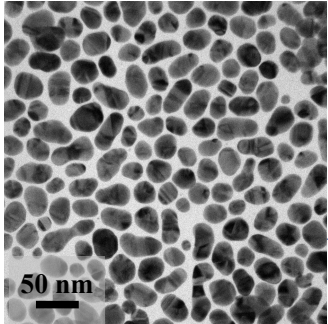
3. SERS for Gas Sensing

SERS substrates validated with nerve agents and vesicants



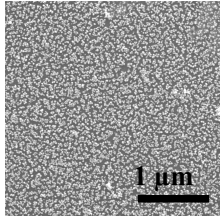
Silver nanoplates electrostatically self-assembled to flat graphite

<https://doi.org/10.1016/j.jhazmat.2019.121279>



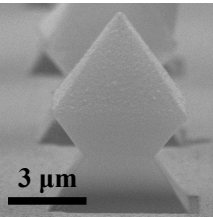
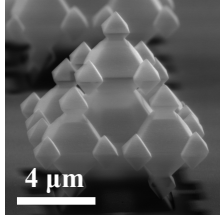
Gold nanoparticles stabilized by citrate ions

<https://doi.org/10.1016/j.snb.2018.04.058>



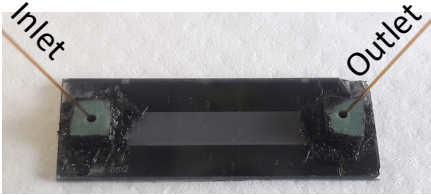
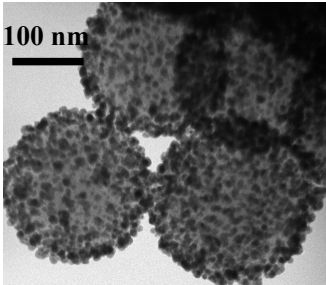
Electrostatically self-assembled... to flat SiO₂

<https://doi.org/10.1016/j.apsusc.2019.144663>



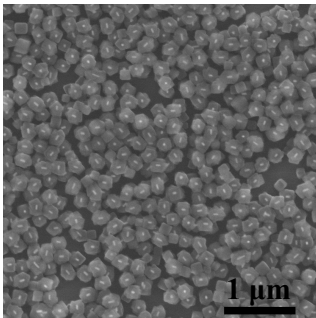
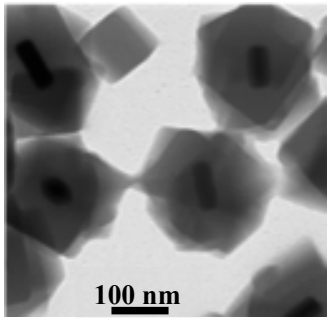
3D fractals

<https://doi.org/10.1007/s00604-020-4216-9>



MCM48@Au nanoparticles spin coated inside Si microfluidic chip

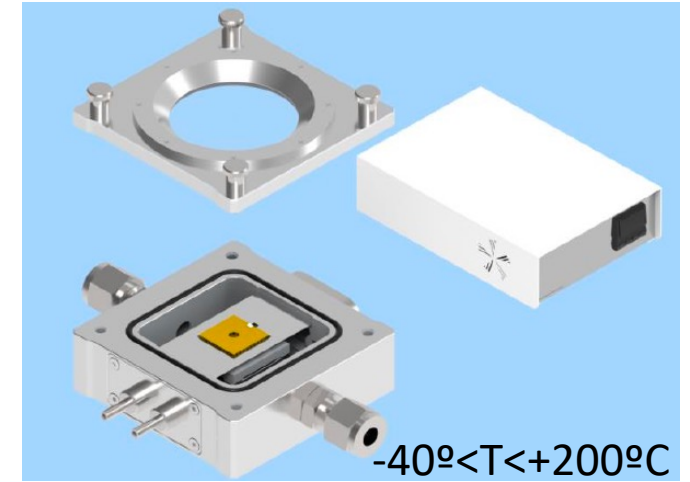
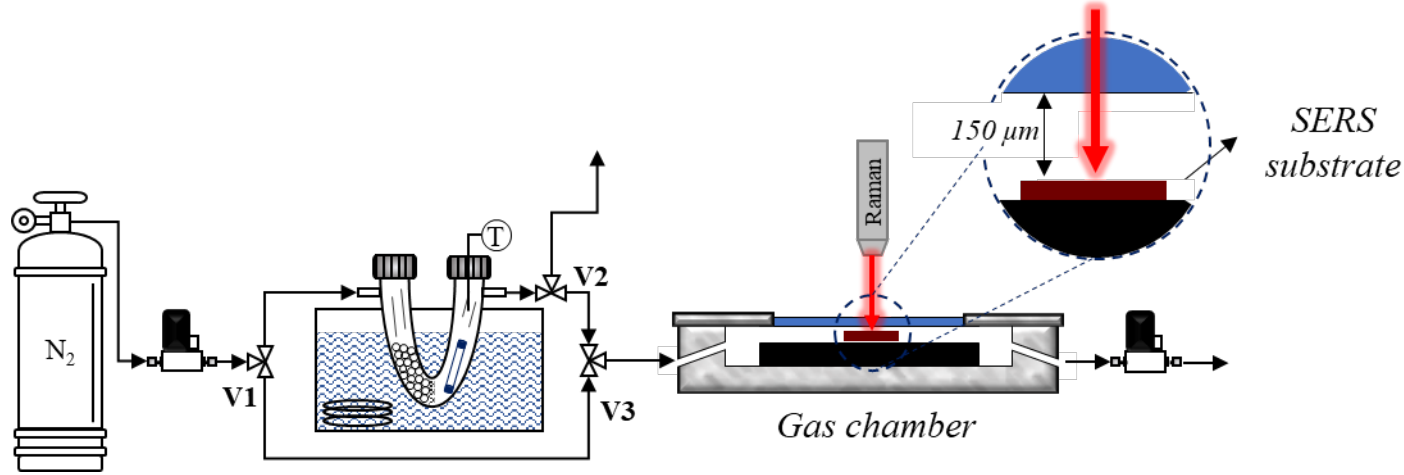
DOI: 10.1109/TRANSDUCERS.2019.8808289



Au@Ag@ZIF8 nanoparticles spin coated onto flat SiO₂

<https://doi.org/10.1021/acssensors.1c00178>

Basic Experimental set-up for SERS in gas phase



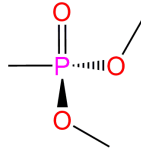
Surrogate/TIC

Real CWA/TIC

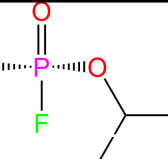
Lab testing*

AEGL1** at 10'

DMMP
(Dimethyl methylphosphonate)



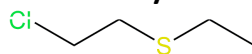
Sarin gas
& G family NAs



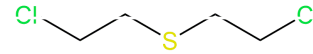
< 2.5 ppmV

1.2 ppbV

CEES
(2-chloroethyl ethyl sulphide)



Vesicant mustard gas



< 100 ppbV

60 ppbV

NH₃ (Ammonia)

Toxic industrial compound

< 30 ppmV

30 ppmV

H₂S (Hydrogen sulfide)

Toxic industrial compound

< 1 ppmV

0.75 ppmV

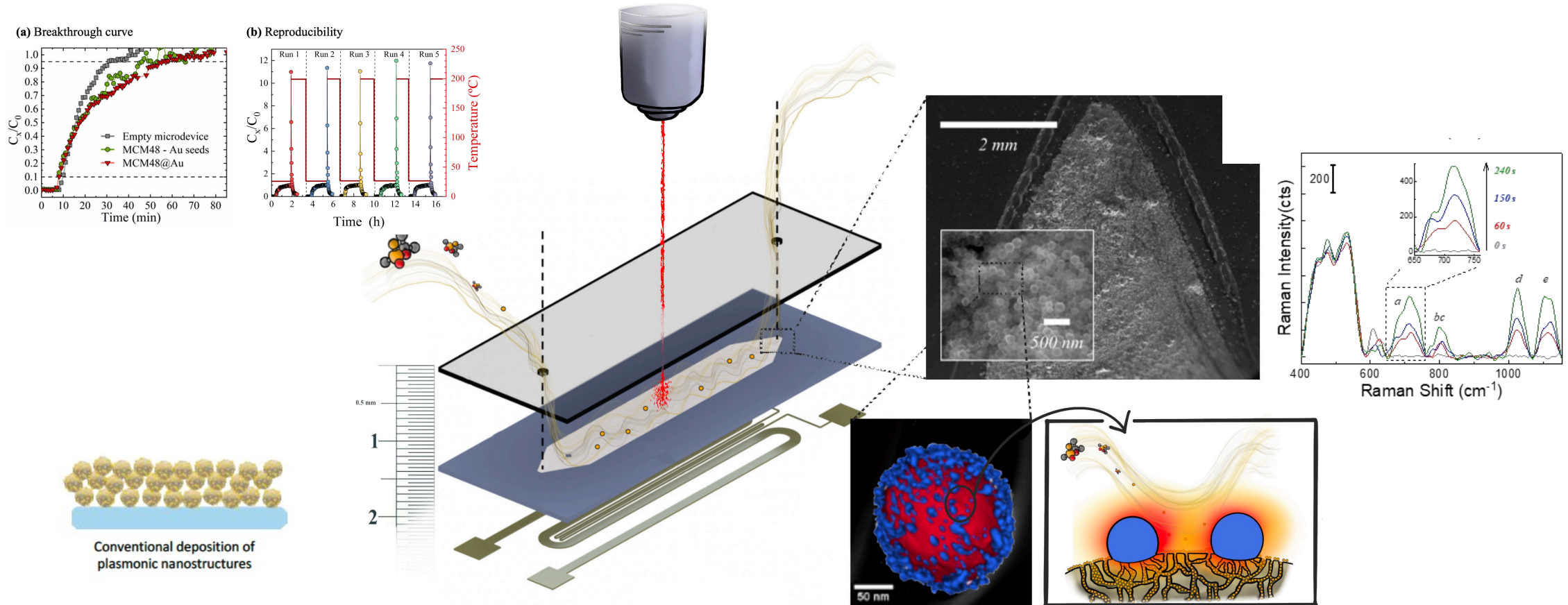
* Concentration values for Metrics definition;

** Irritation and discomfort (non irreversible health effects)

3. SERS for Gas Sensing

Our Approach: From Nanoporous Sorbents to Plasmonic Sorbents

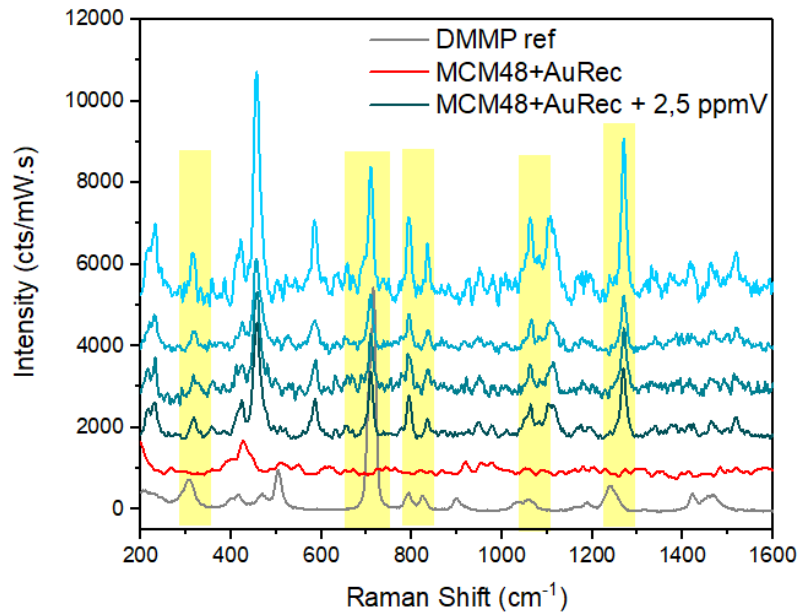
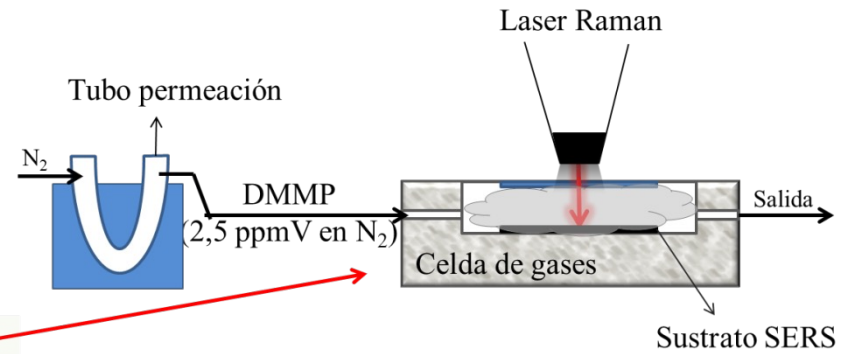
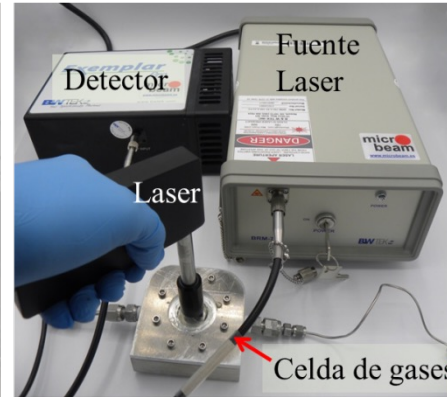
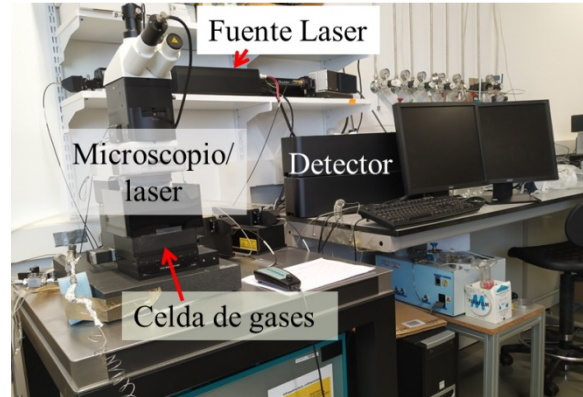
*On-Chip Monitoring of toxic gases: capture and label-free SERS detection with plasmonic mesoporous sorbents**



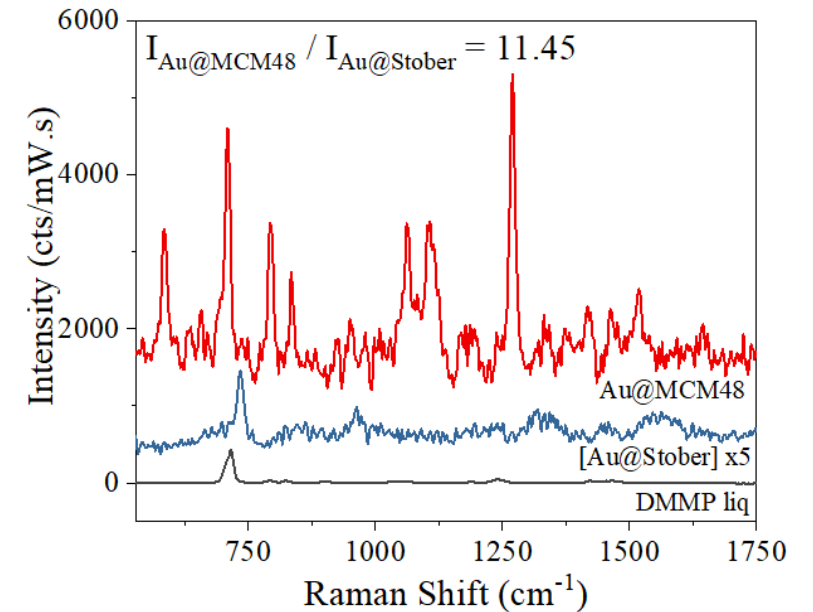
*Manuscript sent to Lab on a Chip Nov 22: Under Revision

3. SERS for Gas Sensing

Our Approach: From Nanoporous Sorbents to Plasmonic Sorbents*



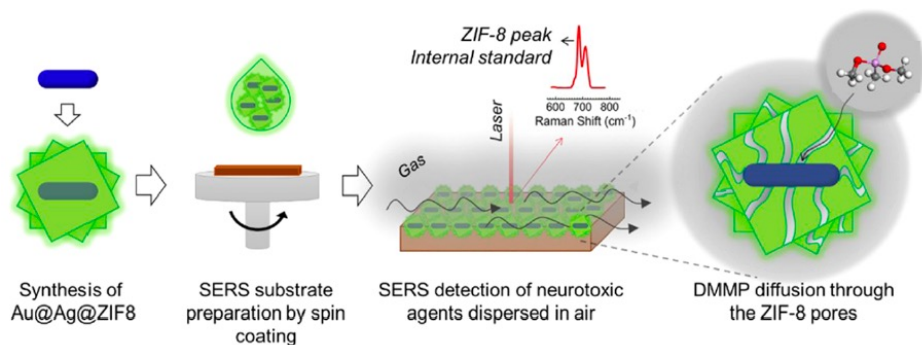
- ✓ Selective Fingerprint of the target molecule: 5 characteristic Raman Bands
- ✓ Outperformance vs. Au@ dense Silica



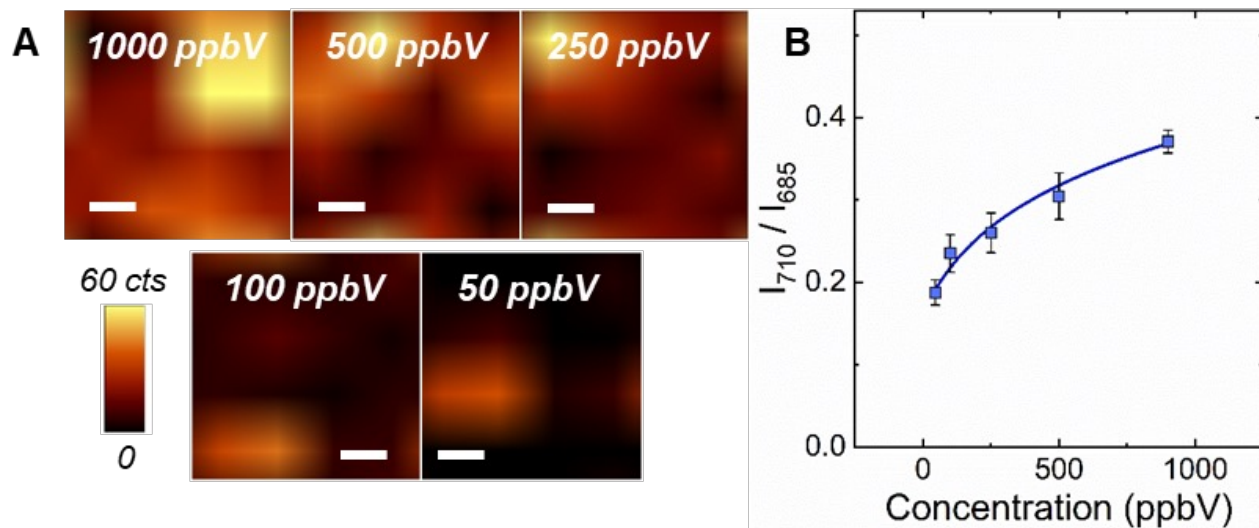
*Manuscript sent to Lab on a Chip Nov 22: Under Revision

3. SERS for Gas Sensing

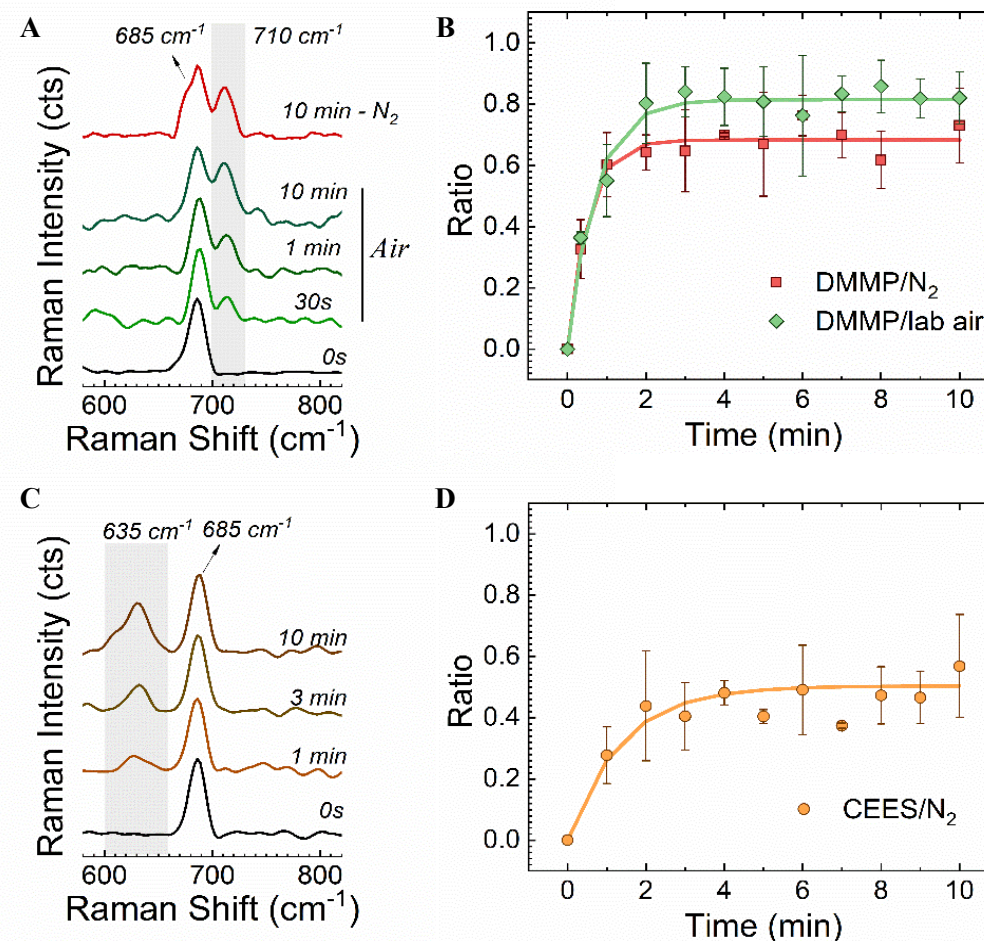
Plasmonic MOF Thin Films with Raman Internal Standard for Fast and Ultrasensitive SERS Detection of Chemical Warfare Agents in Ambient Air*



SERS detection of gaseous DMMP in N₂ using Ag@Au@ZIF-8 thin film as sensing platform: LOD_t=6 ppbV; t_{response,50%}<30s



On-field detection capabilities of Ag@Au@ZIF-8 substrate using a portable Raman equipment (DMMP-2500 ppbV; CEES-76 ppbV)



*ACS Sens. 2021, 6, 2241–2251. <https://doi.org/10.1021/acssensors.1c00178>

3. SERS for Gas Sensing

Future Work

Lack of **miniature**, **fieldable** and **affordable** tools and systems for **detection, identification and monitoring (DIM)** of **Chemical Threats (AEGLs @ ppb level)**



International Forum to Advance
FIRST RESPONDER INNOVATION

- SERS Ad-ons for Near-real time or on-demand detection and identification of chemical threats
- Handheld or robot-mounted instrumentation
- Develop machine learning based algorithms for SERS/Raman data analysis
- Provide data map to aid first responders and incident commanders in hazard assessment and decision-making



4. Take-Home Messages

- Multidisciplinary Field
- Universal and Ultrasensitive Analytical Technique
- Rapid Development of Portable Instrumentation
- SERS Community is devoted to demonstrate performance in Practical Applications, including DIM of traces

Main Areas Contributing to SERS at the Present

